## Radiation Safety Training

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### Outline

Module 1 : Regulations and Licensing
Module 2 : Basic radiation Physics
Module 3 : Radiation and Risks
Module 4 : Detection Instruments
Module 5 : Transport and Handling
Module 6 : Working in Laboratories

MODULE 1 : Regulations and Licensing



Commission canadienne de sûreté nucléaire

#### http://www.cnsc-ccsn.gc.ca

#### The Canadian Nuclear Safety Commission

#### An Overview

Nuclear safety is everyone's business. Every day, millions of Canadians use nuclear energy, though we may not always be aware of how it contributes to our lives.

The Canadian Nuclear Safety Commission (CNSC) regulates the use of nuclear energy and materials to protect health, safety, security and the environment and to respect Canada's international commitments on the peaceful use of nuclear energy. Created in 1946 as the Atomic Energy Control Board, the agency changed its name in 2000 with the enactment of the *Nuclear Safety and Control Act* (NSCA). Its vision is to be one of the best nuclear regulators in the world by being effective, efficient, transparent and an employer of choice.



Canadian Nuclear Safety Commission Commission canadienne de sûreté nucléaire

#### **CNSC Mission**

To regulate the use of nuclear energy and materials to protect health, safety, security and the environment and to respect Canada's international commitments on the peaceful use of nuclear energy.





#### **CNSC Mandate**

Under legislation enacted by Parliament, and policies, directives and international commitments of the federal government, we:

- regulate the development, production and use of nuclear energy in Canada;
- regulate the production, possession, use and transport of nuclear substances, and the production, possession and use of prescribed equipment and prescribed information;
- implement measures respecting international control of the development, production, transport and use of nuclear energy and nuclear substances, including measures respecting the non-proliferation of nuclear weapons and nuclear explosive devices; and
- disseminate scientific, technical and regulatory information concerning the activities of the CNSC, and the effects on the environment, on the health and safety of persons, of the development, production, possession, transport and use of nuclear substances.

#### **Regulatory & Licensing Information**

#### Acts, Regulations and By-laws

The Nuclear Safety and Control Act (NSC Act) came into force on May 31, 2000, when it replaced the Atomic Energy Control Act. It provides the Canadian Nuclear Safety Commission with its regulatory authority.

Under the NSC Act, the Commission has put in place a number of regulations and bylaws. Electronic versions of the key regulations and bylaws administered by the CNSC can be found below.

#### Regulations

- Canadian Nuclear Safety Commission Rules of Procedure (PDF)
- General Nuclear Safety and Control Regulations (PDF)
   O Office Consolidation (PDF)
  - ➡ Radiation Protection Regulations (PDF)
    - Class I Nuclear Facilities Regulations (PDF)
    - Class II Nuclear Facilities and Prescribed Equipment Regulations (PDF)
    - Uranium Mines and Mills Regulations (PDF)
    - Nuclear Substances and Radiation Devices Regulations (PDF)
    - Packaging and Transport of Nuclear Substances Regulations (PDF)
       Office Consolidation (PDF)
    - Nuclear Security Regulations (PDF)
    - Nuclear Non-Proliferation Import and Export Control Regulations (PDF)
    - CNSC Cost Recovery Fees Regulations (HTML) (PDF)

#### Bylaws

Canadian Nuclear Safety Commission By-laws (PDF)

Canadian Nuclear Safety Commission (CNSC)

"The mission of the Canadian Nuclear Safety Commission (CNSC) is to regulate the use of nuclear energy and materials to protect health, safety, security and the environment and to respect Canada's international commitments on the peaceful use of nuclear energy."

The Canadian Nuclear Safety Act

- Defines Nuclear Energy Workers & Nuclear Facilities
- Establishes the Canadian Nuclear Safety Commission (CNSC)
  - Defines duties, powers and responsibilities
  - Ability to make regulations and issue licenses
  - Defines powers of inspectors
  - Defines "opportunity to be heard"
- Establishes penalties for offenses and noncompliance
- Defines Canada's participation in international safeguards treaty

The CNSC is the "Court of Record"

- May issue summons or administer oaths
- Not bound by legal rules
- Its decisions become rules of Federal Court.

 CNSC inspectors may enter and inspect at any reasonable time.

- May search without a warrant
- May order any measures to protect the environment or safety of persons.
- Every person must give reasonable assistance

Offences and Punishments defined in the act.

- Failure to assist inspector (\$5000 or 6 months)
- Misuse of safety equipment
- Failure to comply with license condition, act or any order
- Falsification of records
- Failure to report for duty (!)

Possession of nuclear material that may be used to make nuclear explosive device

(10 years)

#### CNSC Regulations (Rad. Protection)

The CNSC regulations define:

Obligations of licensees and Nuclear Energy Workers (NEWs)
Radiation dose limits
Dosimetry services
Labelling and signs
Records to be kept

### CNSC Regulations (General)

**Obligation of Licensees :** 

Ensure presence of qualified staff

- Train workers to carry out activities in accordance with the act
- Take reasonable precautions to protect environment and health and safety of persons
- Maintain devices within specifications
- Take all reasonable precautions to prevent release of radioactive material

### CNSC Regulations (Rad. Protection)

**Obligations of Licensees :** 

Every licensee shall implement a radiation protection program to:

- Keep the exposures As Low As Reasonably Achievable (ALARA)
- Ascertain the quantity and concentration of release of radioactivity
- Ascertain and record the doses to radiation workers

### CNSC Regulations (General)

#### **Obligation of Workers :**

- Comply with procedures in a responsible and reasonable manner
- Comply with measures established to protect environment and health and safety of persons
- Promptly inform supervisor of:
  - Increase in risk
  - Threat to security
  - Failure to comply with act, regulations or license
  - Any act of sabotage or theft of nuclear substance
  - Unauthorized release of radioactive substance
- Observe and obey all notices and warnings
- Take all reasonable precautions for own safety and of safety of others

A Nuclear Energy Worker

In the Canadian Nuclear Safety Act a nuclear energy worker is defined as:

"a person (worker) who is required, ..., to perform duties in such circumstances that there is a reasonable probability that the person may receive a dose of radiation that is greater than the prescribed limit for the general public."

#### CNSC Regulations (Rad. Protection)

Obligations of Nuclear Energy Workers (NEWs) :

- Every nuclear energy worker shall, on request by the licensee, inform the licensee of the workers:
  - Given names, surname and previous surnames
  - Social Insurance Number
  - Sex
  - Date, province and country of birth
  - Dose record for the previous one-year and fiveyear dosimetry periods

CNSC Regulations (Rad. Protection) Pregnant Nuclear Energy Workers:

Every NEW who becomes aware that she is pregnant shall *immediately* inform the licensee *in writing*.

On being informed by a NEW that she is pregnant, the licensee shall, in order to comply with section 13 (dose limits) make any accommodation what will not occasion costs or business inconvenience constituting undue hardship to the the licensee.

#### The CNSC Regulations define limits on effective dose.

Person	Period	Effective Dose (mSv)
Nuclear Energy Worker	One-year dosimetry period	50
(Including a pregnant NEW)	Five-year dosimetry period	100
Pregnant NEW	Balance of the pregnancy	4
A person who is not a NEW	One calendar year	1

#### The CNSC Regulations define limits on equivalent dose.

Organ or Tissue	Person	Period	Equivalent Dose (mSv)
Lens of an eye	NEW	One-year dosimetry period	150
	Any other person	One calendar year	15
Skin	NEW	One-year dosimetry period	500
	Any other person	One calendar year	50
Hands and feet	NEW	One-year dosimetry period	500
	Any other person	One calendar year	50

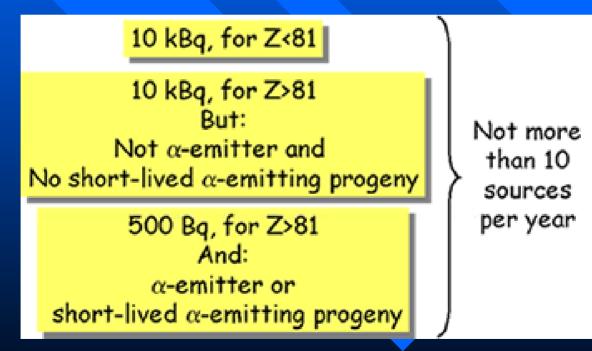
These limits may be exceeded if there is a <u>life</u> threatening emergency, BUT

-Effective dose must remain less than 500 mSv -Equivalent dose (skin) must be less than 5000 mSv



### CNSC Regulations (Nuclear Substances)

Prescribed Nuclear Substances are defined as quantities of radioactivity greater than the Exempt Quantities listed in schedule 1 of the act



Exempted Quantities are not regulated.

#### Exempted Quantities

A license is NOT required to :

Possess, transfer, import, export, use, mine, produce, refine, convert, enrich, process, reprocess, manage or store a nuclear substance, if the quantity of the nuclear substance does not exceed its exemption quantity (NSRDR 5.1)

### CNSC Regulations (Nuclear Substances)

Radioactive material that is used on site is required to have the following documentation:

- Name, quantity, form and location
- Model and serial number (sealed sources)
- Quantity used
- Manner in which it is used
- Name of each worker who uses or handles
- Record of transfer, receipt, disposal or abandonment
- Record of training of each worker (for 3 years)
- Record of all inspections, measurements and tests

### MUHC Licenced Activities

#### Nuclear Substances and Radiation Devices

- Nuclear Medicine (diagnostic)
- Nuclear Medicine (therapeutic)
- Gammacell Irradiator (self-shielded type)
- Consolidated Activities (Research)

#### Class-II Equipments

- Brachytherapy
- Cobalt Teletherapy
- Accelerators
- Calibration

For Nuclear Substances Licensing :

- MUHC has one contact person at the CNSC, J.-C.
   Poirier
- The CNSC has one contact person at the MUHC, C. Janicki, RSO/NSRD.

For Class II Equipment Licensing :

- MUHC has one contact person at the CNSC, A. Licea
- The CNSC has one contact person at the MUHC, M. Evans, RSO/Class II.

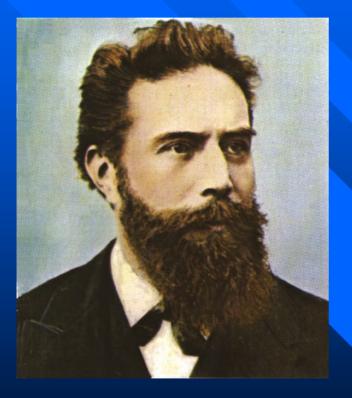
# MODULE 2 Basic radiation physics

# A Brief History

X-rays were first observed in 1895 by A.H. Roentgen

- On the 8th of November, he covers with a black strong paper an apparatus that he uses to study electricity phenomena
- The screen placed nearby seems shining some green light
- His hand placed behind the screen shows the shadow of his handbones!
- At the end of December, he publishes a short article, claiming for a fantastic news: the existence of an unknown and strange radiation, that is thus quickly named "X rays".
- For this discovery, he receives the first physics Nobel price in 1901.

### Roentgen discovers X-Rays



A.H. Roentgen

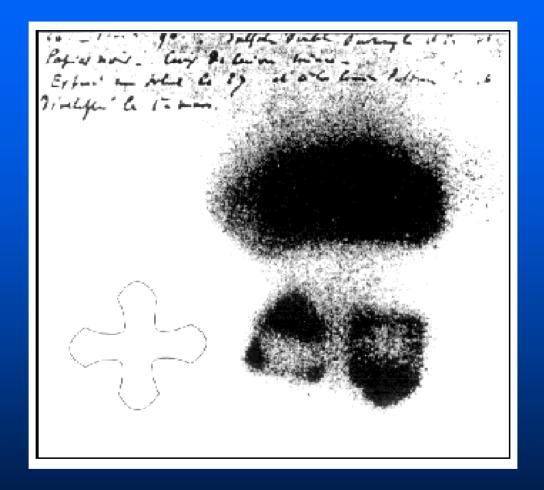
Roentgen's wife hand with wedding ring

# A Brief History

 Radioactivity was first observed in 1896 by A. H. Becquerel

- salts of uranium are brought into the vicinity of an unexposed photographic => the plate becomes exposed
- uranium salts also causes a charged electroscope to discharge
- the salts exhibit phosphorescence and are able to produce fluorescence

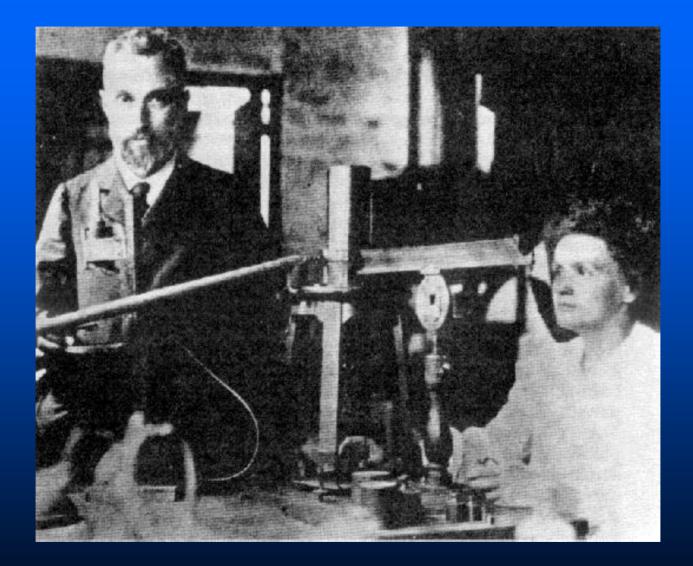




Photographic plate of Becquerel impressed by the radioactivity of uranium.

## A Brief History

- Marie and Pierre Curie extended the work on radioactivity
  - Demonstrate the radioactive properties of thorium
  - Isolate a new material 1 000 000 times more radioactive than uranium : "polonium"
  - In 1898, from penchblende ore (many tons!), they extract by hand some milligrams of an other new material, 2 500 000 times more radioactive than uranium: the "radium"
  - They receive the Nobel price of Physics in 1903 for the discovery of radium
  - Marie Curie received the Nobel price of Chemistry in 1911 for chemistry works on radium

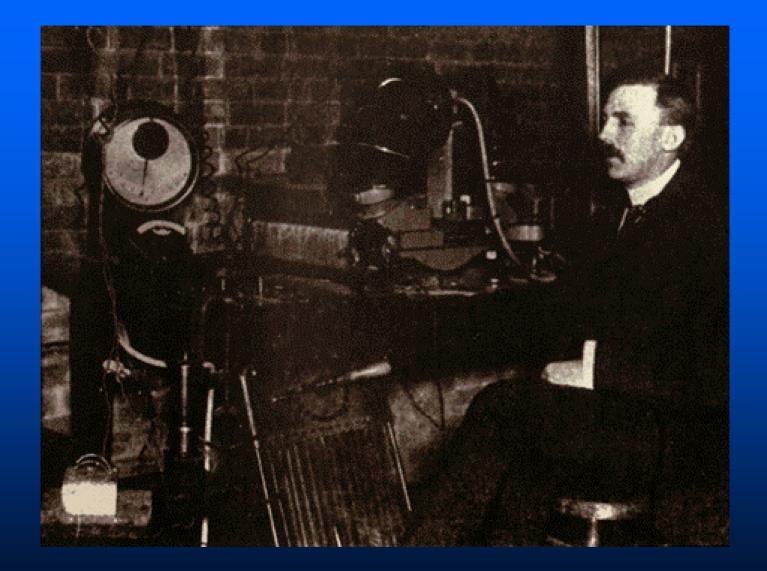


Marie and Pierre Curie

# A Brief History

#### **E.** Rutherford

- Discovers alpha and beta radiation in 1899
- The first to elucidate the related concepts of the half-life and decay constant
- With Frederick Soddy at McGill University, Rutherford showed that elements such as uranium and thorium became different elements (i.e., transmuted) through the process of radioactive decay
- For this work, Rutherford won the 1908 Nobel Prize in chemistry.

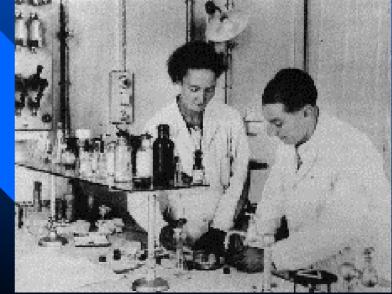


Ernest Rutherford in his Laboratory at McGill University, Ca. 1903

# A Brief History

Frédéric and Irène Joliot-Curie discovered the first example of artificial radioactivity in 1934 by bombarding nonradioactive elements with alpha particles

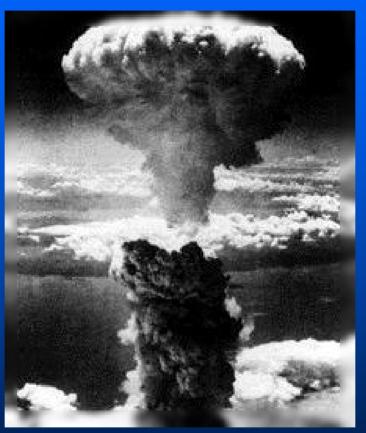
- By shooting an aluminium sheet with alpha particles (helium nuclei), they were able to make radioactive phosphorus



# A Brief History

- On July 4, 1934 Leo Szilard filed a patent application for the atomic bomb
- December 1935, Chadwick won the Nobel Prize for discovery of the neutron.
- January 13, 1939 Otto Frisch observed fission directly by detecting fission fragments in an ionization chamber.
- January 29, 1939 Robert Oppenheimer hears about the discovery of fission and realized that it might be possible to build a bomb.
- Manhattan Project began on September 17, 1942 led by R.O.

#### In 1945, we enter the NUCLEAR AGE



On July 16, 1945, Oppenheimer witnessed the first explosion of an atomic bomb in the New Mexico desert.

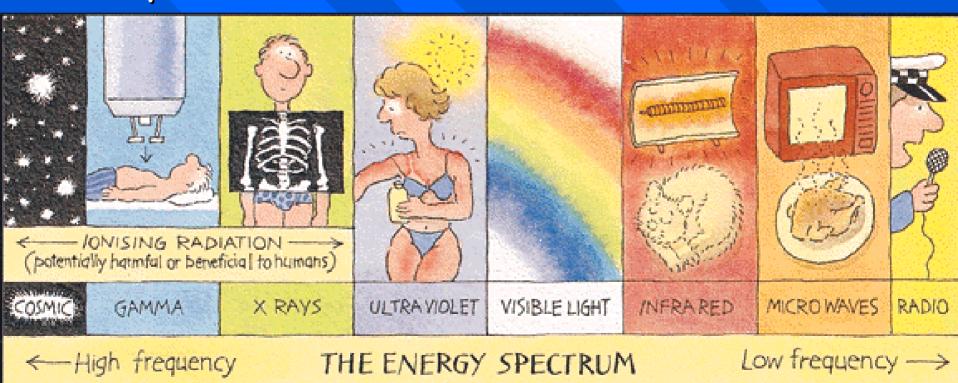
"We knew the world would not be the same" R.O.

## What is Radiation?





#### Radiation: The transmission of energy through space in the form of waves or streams of particles.



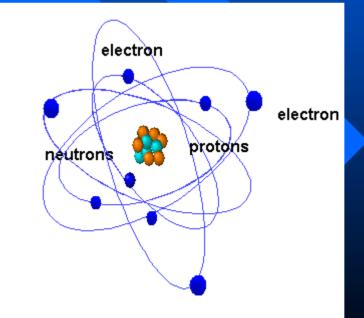
# What is Ionizing Radiation?



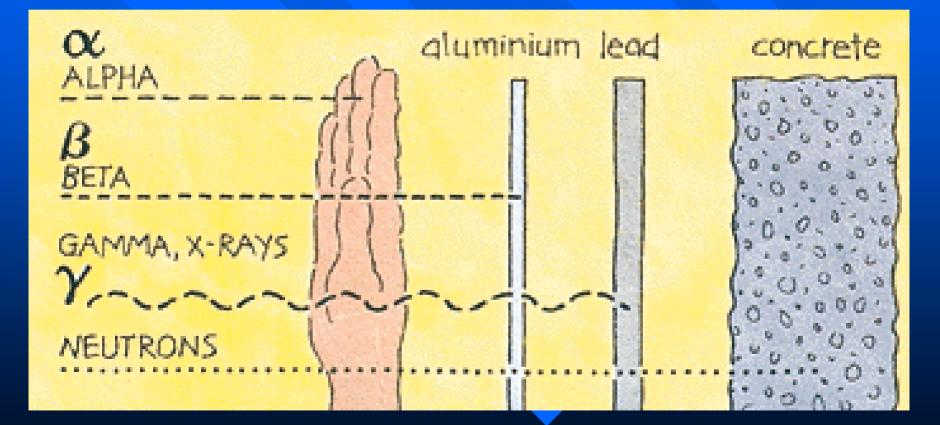


# Ionizing Radiation

Radiation that has sufficient energy to ionize, or remove electrons from, an atom.



# Types of Ionizing radiation

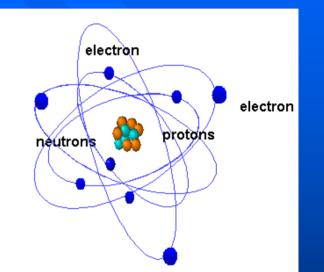


# Types of Ionizing radiation

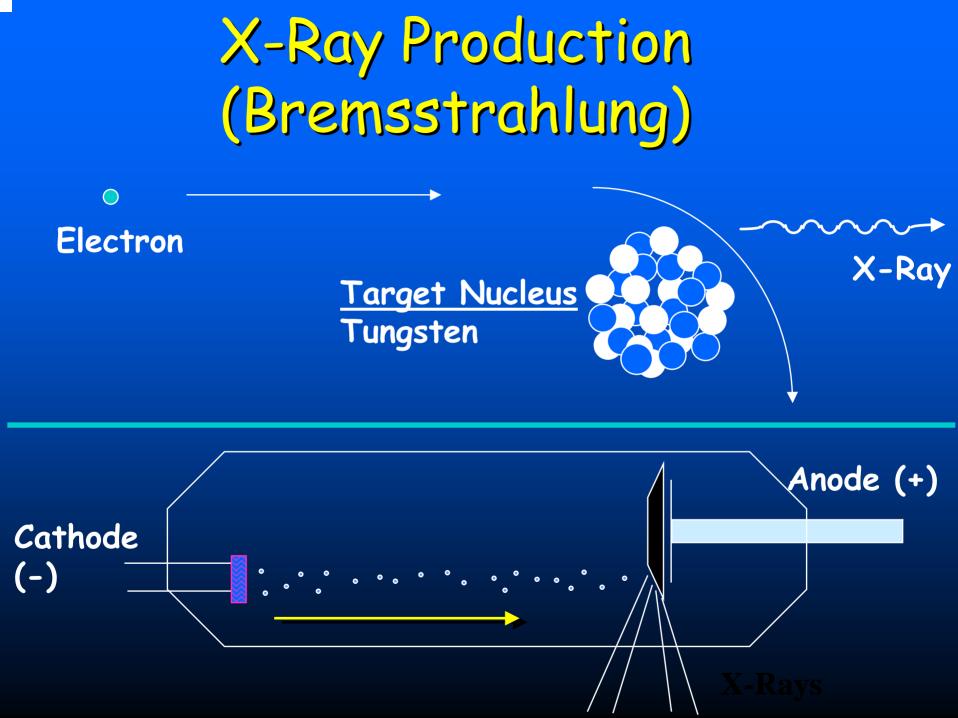
Orbital Electrons
 - X-Ray emission

#### Nuclear Decay

- gamma emission

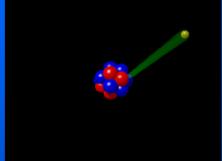


- beta (electron, positron) emission
- alpha particles
- neutrons



Nuclear Decay and Radioactivity

WHAT IS NUCLEAR DECAY ?



spontaneous transformation of an unstable combination of nucleons (PARENT) to a less unstable combination of nucleons (DAUGHTER)

#### WHAT IS RADIOACTIVITY ?

spontaneous emission of particulate and/or electro-magnetic radiation

## Units of Radioactivity

Radioactivity is measured in Beguerels (Bg)

- ·1 bequerel (1Bq)

= 1 transformation/second 1 megabequerel (1MBq) = 10<sup>6</sup> transformations/sec
 1 gigabequerel (1GBq) = 10<sup>9</sup> transformations/ sec

The Beguerel has recently replaced the Curie as the SI unit of radioactivity:

1 curie = number of transformations/sec/g radium =  $3.7 \times 10^{10}$  transformations/second 37 GBg (large!) Ξ

## Decay and Half-lifes

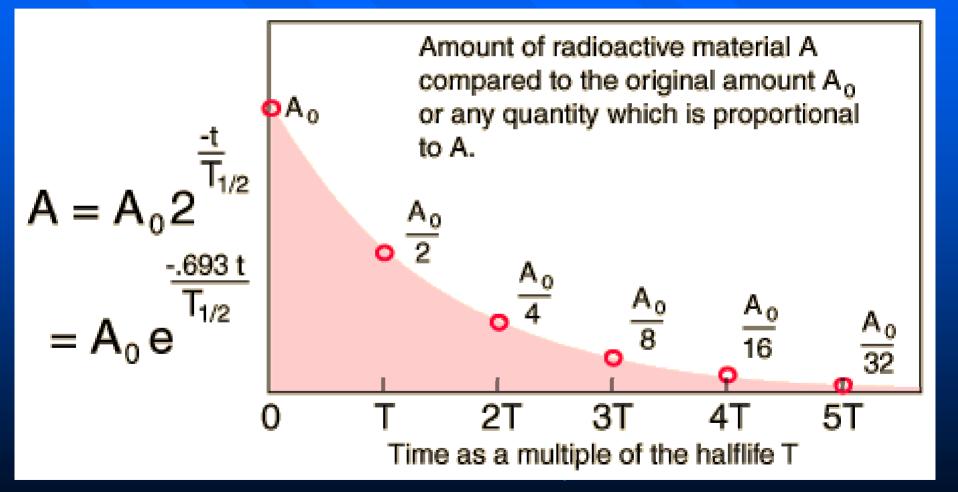
Radioactive decay is governed by laws of probability (quantum mechanics)

Nucleus X will "probably" decay during time interval *At* with probability *p(At)* 

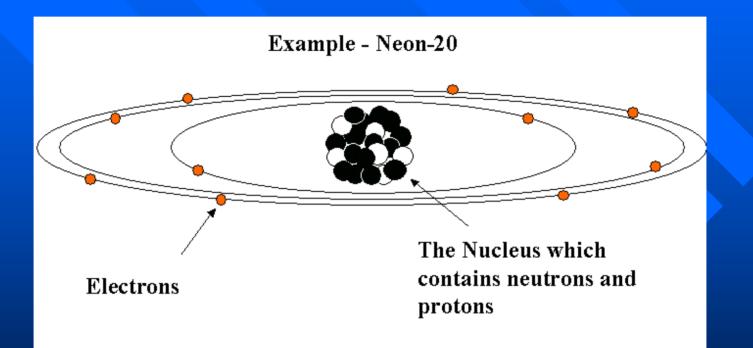
For  $\Delta t$  infinite, p = 1

- The time for p = 0.5 is the half-life  $T_{\frac{1}{2}}$ 

## Decay and Half-lifes



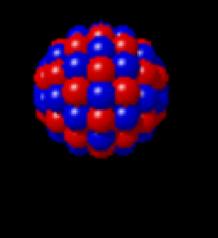
# Basic Nuclear Theory

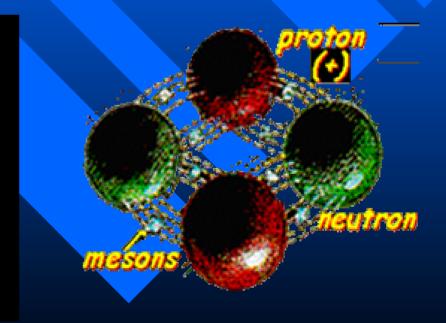


An Atom contains Protons, Neutrons and Electrons

# Basic Nuclear Theory The nucleus is composed of neutrons (n) and protons (p) n and p are bound together by mesons

Protons have a positive charge neutrons have no charge





# Basic Nuclear Theory

The number of protons in the nucleus is known as the atoms atomic number Z

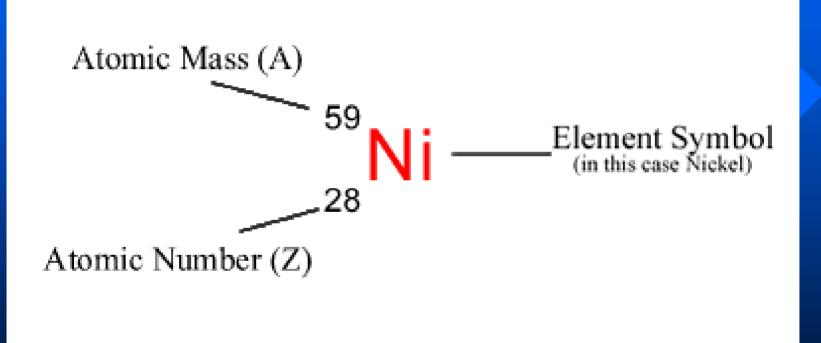
The total number of particles (p + n) in the nucleus is called the atoms **atomic mass** A

Example : He atom

Z = 2 N = 2 A = N + Z = 4



## Notation



# Chart of the Nuclides

Half life Stable													38Ti	<sup>39</sup> Ti	40Ti	⁴¹Ti	42Ti	43Ti	44Ti	⁴5Ti	46Ti
	<ul> <li>Very short</li> <li>&gt; 100,000 yr</li> <li>&gt; 10 yr</li> <li>&gt; 10 yr</li> <li>&gt; 100 days</li> </ul>									<sup>36</sup> SC	37SC	<sup>38</sup> Sc	39SC	40SC	41SC	42 <mark>8</mark> C	43Sc	44 <mark>Sc</mark>	45 <mark>8</mark> 0		
										<sup>35</sup> Ca	<sup>36</sup> Ca	з7Са	38Ca	<sup>39</sup> Ca	40Ca	41Ca	⁴²Ca	₄₃Ca	⁴⁴Ca		
	> 10 days > 1 day > 1 hr > 1 min. 24P 22Si 23Si 21Al 22Al 19Mg 20Mg 21Mg			32Қ				ззК	з4К	зъК	зеК	37К	звК	зәҚ	40K	41K	42K	₄зК			
				30				31År	32År	зздг	34Ar	з5Ar	зедг	<sup>37</sup> År	задг	зэдr	40År	41År	42Ar		
						28C]	29Cl	30Cl	31Cl	32C1	33C]	34( <mark>1</mark>	35CI	36C]	37Cl	38 <mark>(</mark> ]]	39CJ	40Cl	₄ıCl		
						26S	27S	28S	29S	30S	31S	32S	ззд	34S	35S	36S	37S	38S	39S	40S	
						24P	25P	26P	27P	28P	29P	зөр	31P	32р	ззр	34р	35р	зер	37р	звр	зэр
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				<sup>18</sup> Na	19Na	<sup>20</sup> Na	<sup>21</sup> Na	<sup>22</sup> Na	<sup>23</sup> Na	<sup>24</sup> Na	25Na	26Na	27Na	<sup>28</sup> Na	29Na	<sup>30</sup> Na	<sup>31</sup> Na	<sup>32</sup> Na	<sup>33</sup> Na	34Na	35Na
				17Ne	<sup>18</sup> Ne	19Ne	20 <mark>1</mark> 1e	21Ne	<sup>22</sup> Ne	23Ne	<sup>24</sup> Ne	25Ne	26Ne	27Ne	28Ne	29Ne	зø∖уе	зıNe	32[Ve		•
		14F	15F	16F	17F	18F	19F	20F	21F	22F	23F	24F	25F	26F	27F	28F	29F			-	
	12()	13()	14()	150	160	17()	18()	19()	20()	21()	22()	23()	24()	25()	26()						

Ζ

## Nuclear Families

#### Isotopes : Same number of protons Z Different number of neutrons N

Z 8	O -218.79° -182.95° -215.9994 0.078%	016 0+	017 1/2+	018 0+	019 4 s 3/2+ β <sup>.</sup>	O20 8 m 0+ β·	O21 800 ms 3/2+ β·	O22 10 s 0+ β <sup>-</sup>	O23 300 ms 1/2+ β <sup>.</sup>
	7	8	9	10 N	11 J	12	13	14	15

## Nuclear Families

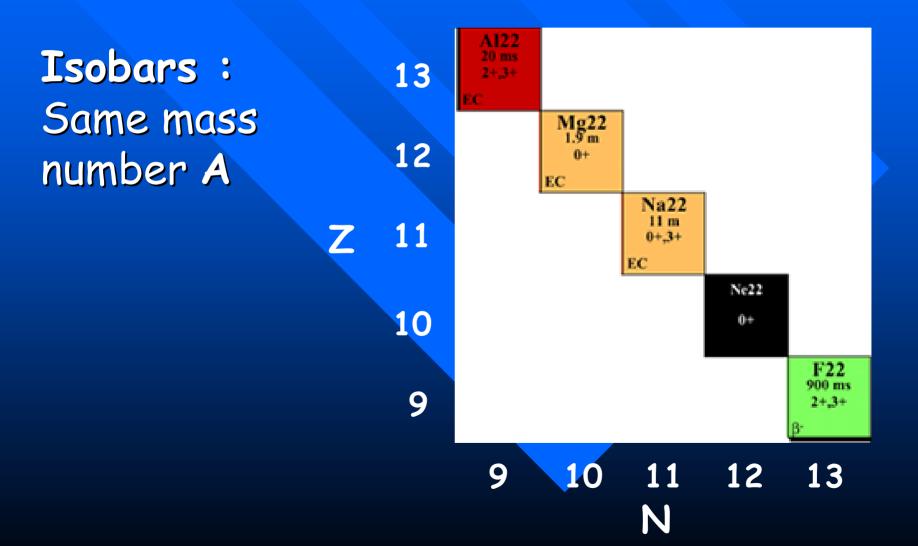
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Isotones : Same neutron number N



9 N

## Nuclear Families



# Line of Stability

Long-lived nuclides are close to the line of stability

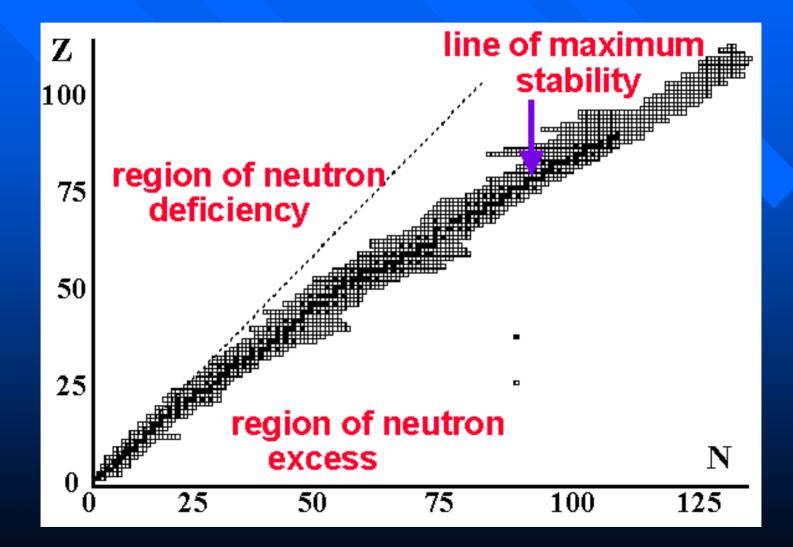
The half-life gets shorter the further away from the line

Excess Proton => above the line

Excess neutron => below the line

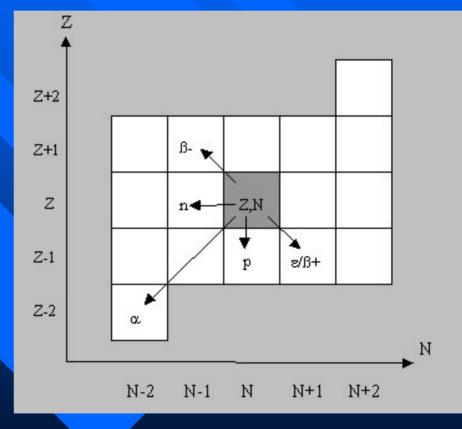
15	P 44.00 40.95 30.973382	P23 9 mil 1/2+	P24 7 2-3-	P25 34 mil 1/2+	P26	P27 100 mm 102 *	P28 110 ms 0+,1+	P29 51 121	P30 185 95,15
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9	F 411-62 -125-62 -125-62 -135-	F17 0 = 10	F18 4 h 0+,1+ EC	F19 12+	F20 131 1-34	F21 181 101	F22 900 ms 2+3+	F23 200 ms 52+	F24 200 mi 1020
8	0 (15.9694 (15.9694 0.0795)	016 #=	017 82-	018 0+	019 41 32- 8	020 5 =	021 800 ms 3/2+	022 185 84	023 38 m 12
		8		10		12		14	

# Line of Stability

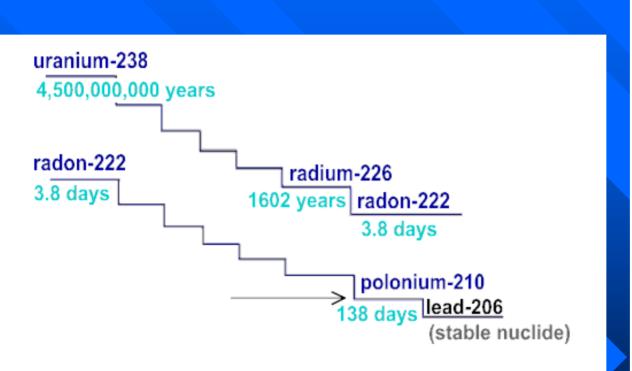


# Types of Nuclear Decay

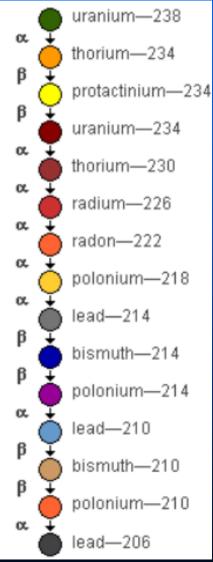
Excess p & n  $\Rightarrow$  alpha Excess p  $\Rightarrow$  positron ( $\beta^+$ ) Excess n  $\Rightarrow$  negatron ( $\beta^-$ ) Excess nuclear E  $\Rightarrow$  gamma



## U-238 decay chain

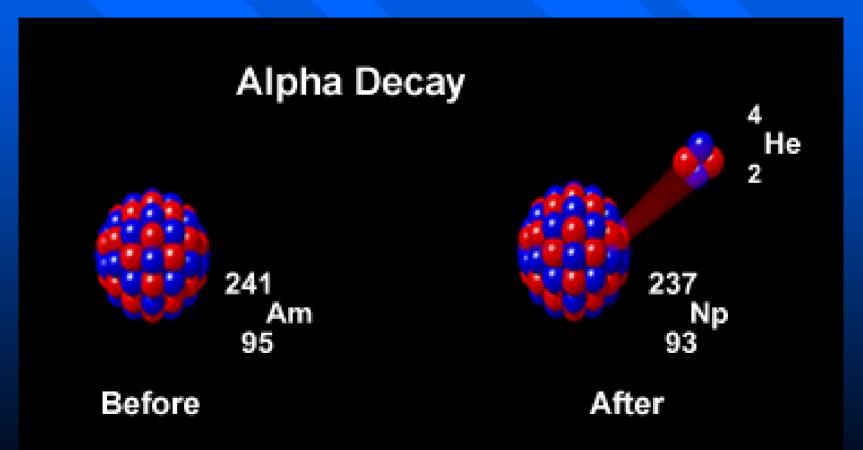


U-238 decays through a series of steps to become a stable form of lead.



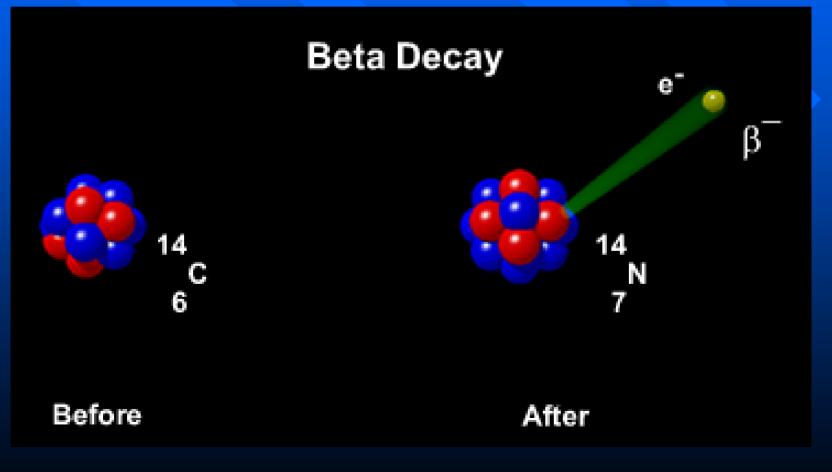
#### ALPHA DECAY

Very large nuclei may need to emit more than one particle in order to become stable. These atoms will emit an alpha particle.



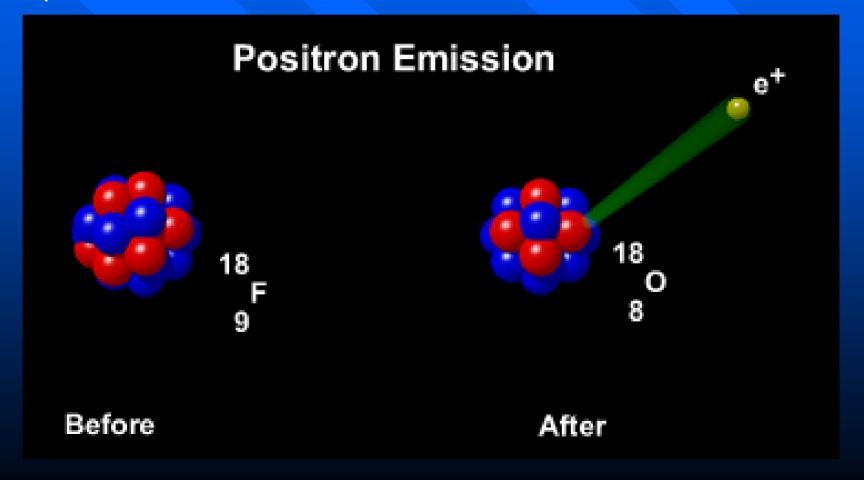
#### BETA DECAY

# Too many neutrons results in decay by the emission of an electron



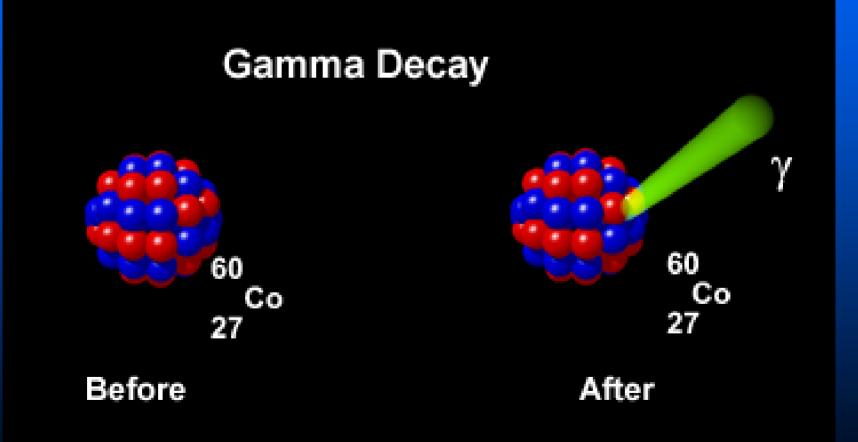
#### BETA DECAY

Too little neutrons results in decay by the emission of a positron



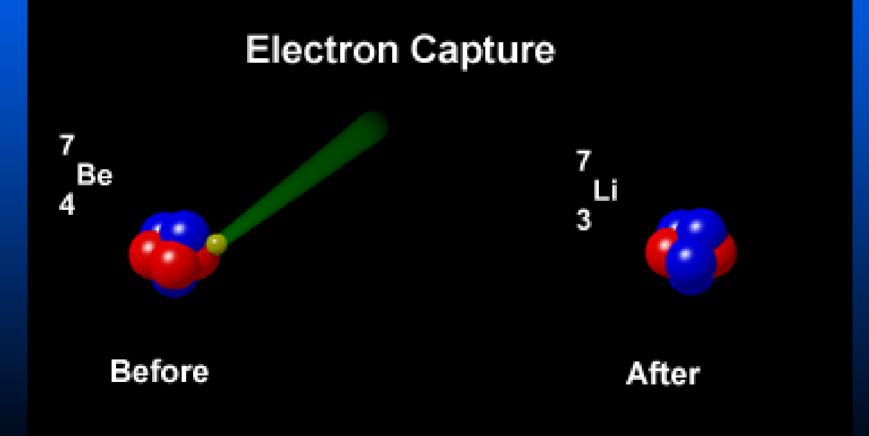
#### GAMMA DECAY

The nucleus may have excess energy that will be released in the form of a gamma ray (  $\gamma$  )



#### ELECTRON CAPTURE

An electron is captured by the nucleus changing one of the protons into a neutron. This causes a decrease in atomic number (Z) but no change to the atomic mass (A).



#### ALPHA EMISSION

- origin: DISINTEGRATING NUCLEUS
- form of radiation: PARTICLE
- energy range: 4-8 MeV
- range of travel: 2-8 cm in air
- other characteristics: LARGE MASS, DOUBLE CHARGE, HIGH SPECIFIC ACTIVITY

#### BETA EMISSION

origin: DISINTEGRATING NUCLEUS

 form of radiation: NEGATRON (electron)
 POSITRON (similar to an electron but positive charge)

energy range:

0.02 - 4.8 MeV

range of travel: 0

0 - 10 m in air

other characteristics:

DIFFERS FROM AN ELECTON IN ORIGIN AND ENERGY, TRAVELS ALMOST THE SPEED OF LIGHT, MASS (9.1× 10<sup>-31</sup> kg)

#### GAMMA EMISSION

- origin: NUCLEUS
- form of radiation: ELECTROMAGNETIC RADIATION (emr - photon)
- energy range: 10 keV 3 MeV
- range of travel: > 100 m in air
- other characteristics: Zero mass, no charge

#### X- RAY EMISSION

- origin: ORBITAL ELECTRON
- form of radiation: ELECTROMAGNETIC RADIATION (emr - photon)
- energy range: 10eV 1 MeV
- range of travel: 100 m in air
- other characteristics: Zero mass, no charge

#### Neutron and Other Radiations

Neutron radiation are produced in nuclear reactors (nuclear fission)
 Neutron can be produced by mixing alpha emitting radionuclide with beryllium

#### $^{9}Be + ^{4}He = {}^{12}C + n$

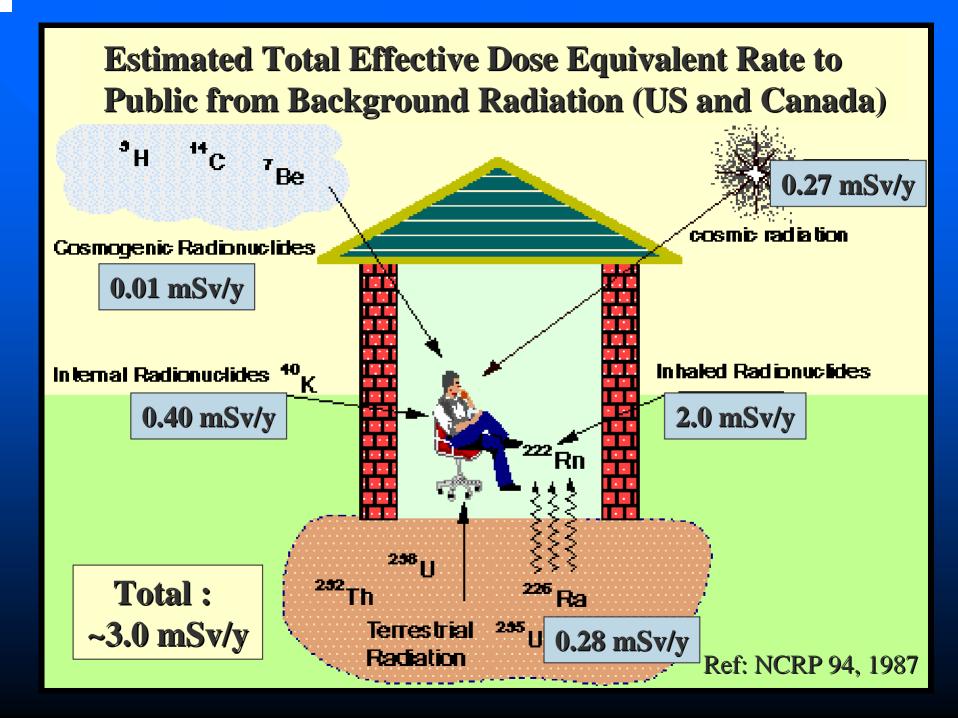
 Other atomic particles (leptons, baryons, mesons) can be produced using particle accelerators (high energy physics)

# Isotope Emissions

Isotopes	Half-life	Type(s)of emission
P32	14.3 d	β (1.709 MeV)
<b>P</b> 33	25.4 d	β (249 keV)
Cr <sup>51</sup>	27.7 d	γ (0.323 MeV, 5 keV)
Fe <sup>59</sup>	44.5 d	β (1.3 MeV, 1.6 MeV)
<b>I</b> <sup>125</sup>	59.6 d	γ(35keV)+ X(27 - 32keV)
<b>S</b> <sup>35</sup>	87.4 d	β (0.167 MeV)
<i>C</i> a <sup>45</sup>	163 d	<mark>β (0.</mark> 257 MeV)
$H^3$	12.4 yr	β (18.6 keV)
<i>C</i> <sup>14</sup>	5760 yr	β (0.156 MeV)

# MODULE 3 Radiation and Risks

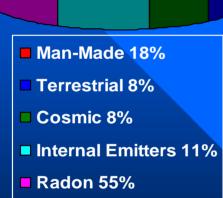




## Sources of Radiation

#### Natural Sources

#### Man Made Sources



The average Canadian receives an annual dose of 3.6 mSv of exposure annually Nuclear Medicine 4%

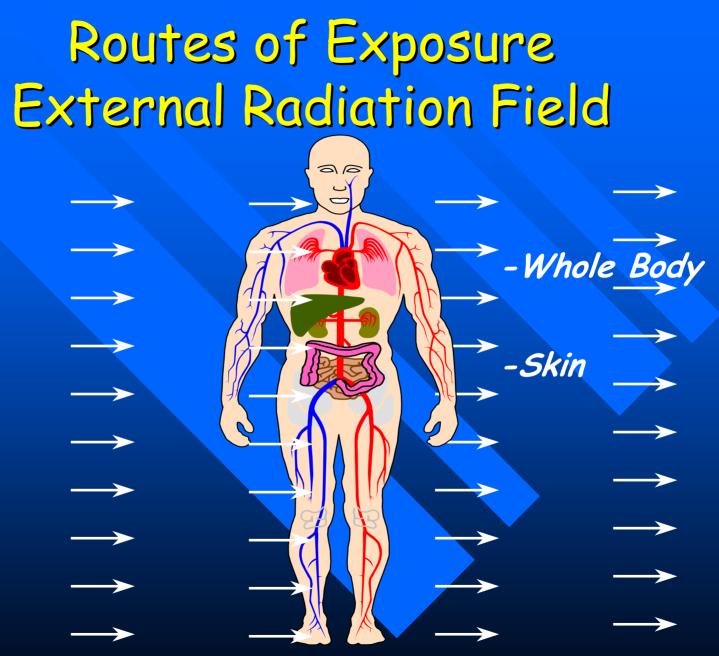
Consumer Products 3%

Other <1%</p>

Medical X-rays 11%

#### Other:

Occupational	0.3%
Fallout	0.3%
Nuclear Power	0.1%
Miscellaneous	0.1%





## Routes of Exposure Internal Contamination

30

#### Absorption through intact skin

Inhalation

Absorption through wound

Ingestion

# X-rays and Matter

3 main interactions for X-rays

Photoeffect
Compton Scattering
Pair Production

## Photosffect

#### All the energy of an X-ray is given to an orbital electron

X-ray



### Result: High energy electron

### Compton Scatter

#### X-ray gives part of its energy to orbital electron

WWWW X-ray



NNN

### Result: High energy electron

## Pair Production

#### X-ray has enough energy to convert into electron-positron pair

X-ray



### Result: High energy electron & positron

### Charged Particles in Matter

High energy electrons transfer their energy to orbital electrons as they move through media

High energy e

Creation of many ions

### Net Results of Interactions

Large number of ionized atoms from electron interactions

High energy electrons

X-ray

Low energy electrons + ions

# The Absorbed Dose D<sub>T,R</sub>

The Absorbed dose (D<sub>T,R</sub> in Gy) is the quotient obtained by dividing the energy absorbed through exposure to radiation by the mass of the body that absorbs that radiation

Units : gray (Gy) = 1 Joule / Kg

# The Equivalent Dose H<sub>T</sub>

The sum of the products, in Sievert (Sv), obtained by multiplying the absorbed dose (D<sub>T,R</sub> in Gy) by the radiation weighting factors (w<sub>R</sub>) for a type of radiation

$$H_T = \Sigma w_R D_{T,R}$$

Type of radiation	WR
Photons	1
Electrons	1
Neutrons	5-20
Protons	5
Alpha	20

# The Effective Dose E

The sum of the products, in Sievert (Sv), obtained by multiplying the equivalent dose (in Sv) received by an organ by the tissue weighting factors W<sub>T</sub> for that organ

### $E = \Sigma W_T H_T$

The tissue weighting factors  $W_T$  is related to the risk of carcinogenesis in the tissue T

# The Committed Dose H<sub>50</sub>

For radioactive material taken into the body, the committed dose is the equivalent dose rate H(t) to an organ integrated over a period of 50 years following intake

$$H_{50} = \int \dot{H}(t) dt$$

# Effective Dose Limits

Person	Period	Effective Dose (mSv)
including a pregnant	<ul> <li>(a) One-year dosimetry period</li> <li>(b) Five-year dosimetry</li> </ul>	50
	period	100
Pregnant nuclear	Balance of the	
energy worker	pregnancy	4
A person who is not a nuclear energy worker	One calendar year	1

# Equivalent Dose Limits

Organ or Tissue	Person	Period	Equivalent Dose (mSv)
Lens of an eye	(a) Nuclear energy worker	One-year dosimetry	
		period	150
	(b) Any other	One calendar	
	person	year	15
Skin	(a) Nuclear energy worker	One-year dosimetry	
	(b) Any other	period One calendar	500
	person	year	50
Hands and feet	(a) Nuclear energy worker	One-year dosimetry	
		period	500
	(b) Any other	One calendar	
	person	year	50

## **Biological Effects of Radiation**

Injury to living tissue results from the transfer of energy to atoms and molecules (ionization)

- Produce free radicals.
- Break chemical bonds.
- Damage molecules that regulate vital cell processes (e.g. DNA, RNA, proteins).

The cell can repair certain levels of cell damage.
 At low doses cellular damage is rapidly repaired.
 At higher levels, cell death results and tissues may fail to function.

## **Biological Effects of Radiation**

#### Deterministic Effects

- Loss of function of tissues in organs due to cell loss (ex. Radiation induced cataracts)
- Result from high dose exposure for which there is a threshold

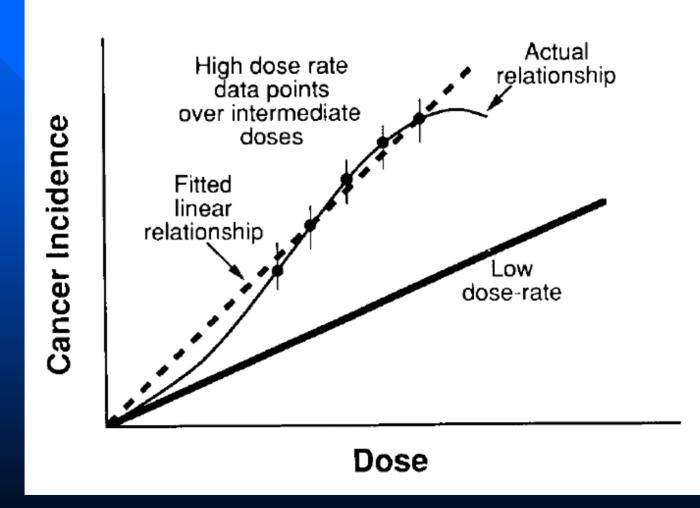
#### Stochastic Effects

- Cell changes (mutation) may occur due to radiation
- Severity of condition is not increased with dose, but occurrence increases
- We assume there is no threshold dose (eg. Cancer, genetic mutation)

# Human Data for Radiation Effects

Japanese survivors of A-bomb Early radiotherapy studies » Britain 1935-44 ankylosing spondytlitis treatments with radiation to spine » Radiology prior to 1922 » 1950's treatment of ringworm in the scalp of children High doses in early diagnostic work » Tuberculosis studies using fluoroscopy, Canada & US

# Linear quadratic Model



#### ICRP and NCRP preferred model

# Linear quadratic Model

Table 19-3. ICRP Summary of Risks of Cancer Lethality by Radiation

	HIGH DOSE HIGH DOSE RATE	LOW DOSE LOW DOSE RATE
Working population	$8 \times 10^{-2}$ per Sv	4 × 10 <sup>-2</sup> per Sv
Whole population	$10  imes 10^{-2}$ per Sv	$5  imes 10^{-2}$ per Sv

(International Commission on Radiological Protection: Recommendations. Annals of the ICRP Publication 60. Oxford, Pergamon Press, 1990)

# Putting Risk into Perspective

Health Risk	Estimated Life Expectancy Lost
Smoking 20 cigarettes a day	6 years
Overweight by 15%	2 years
Alcohol (US Average)	1 year
All accidents	207 days
All natural hazards	7 days
Occupational dose of 3 mSv/year	15 days

Ref. : B. L. Cohen and L. S. Lee, "Catalogue of Risks Extended and Updates," Health Physics, Vol. 61, September 1991.

# Acute Exposure

#### Effects of Acute Exposure to Specific Organs

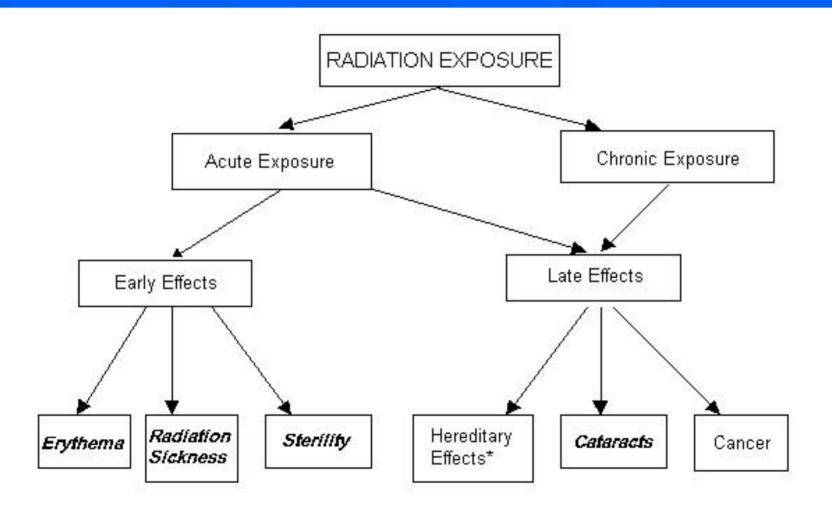
Dose (mSv)	Organ	Effect
3 500	Testes	Permanent sterility
3 500	Eye	Later cataract formation
3 000	Ovaries	Sterility
2 500+	Skin	Skin reddening (erythema) and possible permanent hair loss
500	Bone marrow	Reduced blood cell formation
150+	Testes	Temporary sterility
60	Foetus	Probable minimum dose causing effect (possible malformation)

# Acute Exposure

#### Effects of Acute Whole Body Exposure to Radiation

Dose (mSv)	Effect
50 000+	Severe damage to central nervous system - rapidly lethal.
8 000 - 50 000	Destruction of lining of intestine and white blood cells - death within two weeks.
4 000	Fatal for half those irradiated within 30 days without medical treatment.
2 000 - 8 000	Damage to white blood cells and gut lining. Death may result from secondary infection but can be avoided in many cases with special medical treatment.
1 000 - 2 000	Possible radiation sickness - nausea, vomiting, diarrhoea - not lethal.

# Summary of Radiation Effects



\* Never observed in humans

# ALARA Principle

Doses are required to be kept <u>As Low As</u> <u>Reasonably</u> <u>A</u>chievable

Further ALARA analysis not required if

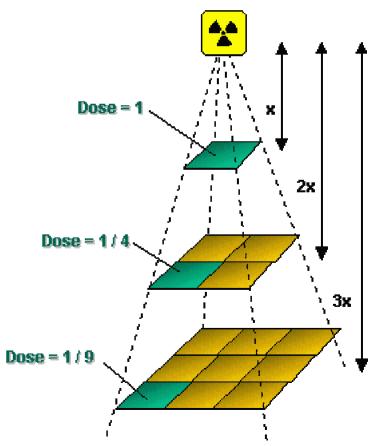
- Individual occupational doses unlikely to exceed 1 mSv/year
- Dose to public unlikely to exceed 50 µSv/year
- Annual collective dose unlikely to exceed 1 person-Sv

Expenditure in excess of \$100,000 to reduce collective dose by 1 person-Sv not justified

### What to do to protect yourself....

#### Basic Radiation Safety Principles for External Radiation



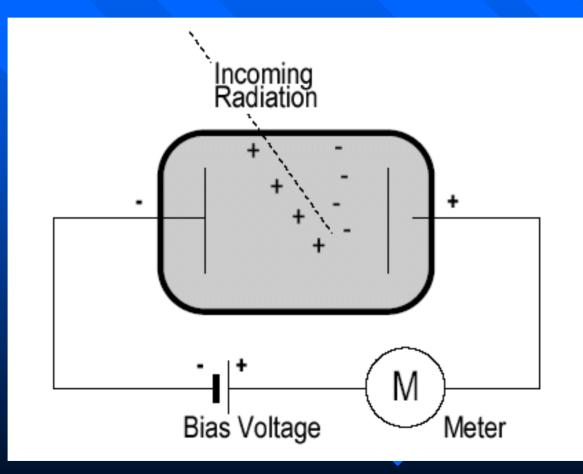


# MODULE 4

Detection Instruments and Survey techniques

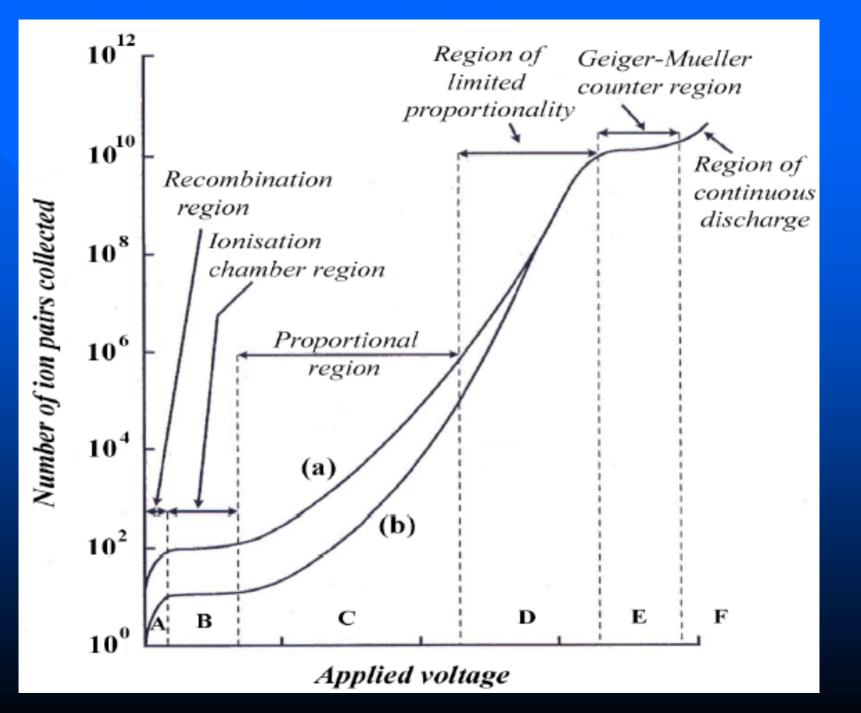
### Types of Radiation Detectors Gas-filled detectors - Ionization chamber - Proportional counter - Geiger-Mueller detector Semiconductor detectors Scintillation detectors - Na(TI)- Liquid scintillation detectors Personal Dose Monitors - TLD, Film badges, EPD

# Gas-Filled Detectors



# Gas-Filled Detectors





## Gas-filled Detectors

Ionization Chamber (B)

- # primary ions collected ~ E deposited
- Use build-up cap at high energy

Proportional counter (C)

- # ions collected ~ E deposited
- Charge amplification of 10<sup>3</sup> -10<sup>4</sup>
- Suitable for low intensity rad. field

## Gas-filled Detectors

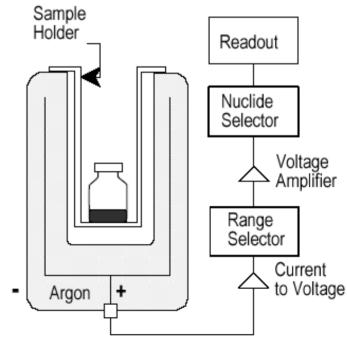
Geiger Mueller detector (E)

- Signal independent of E deposited
- Large amplification of 10<sup>9</sup> -10<sup>10</sup>
- Used at VERY LOW radiation levels
- Indicator of radiation

# Dose Calibrator

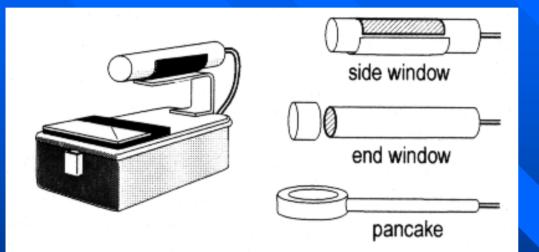
Ionization Chamber (well chamber) Calibration factors for each specific nuclides Measures "Activity" in MBq (or mCi) and not "Dose"





Re-entrant type Dose Calibrator

## The Geiger Mueller Detector



**Portable G-M Monitor** 

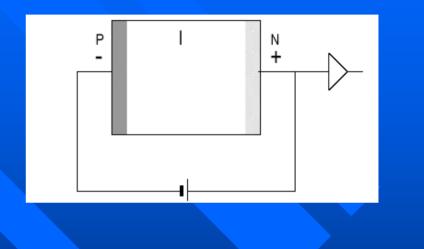


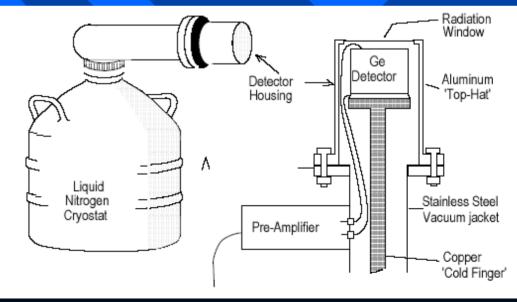
Pancake probe probe (Thin mylar window)

Tritium cannot be detected
Efficiency to C-14 and S-35 is low (<5%)</li>
Efficiency to P-32 is high (~40%)
Efficiency to gamma contamination is low (1-2%)

#### Semiconductor Detectors

- May be viewed as "solid state" ion chambers (P-I-N junction)
- Sensitivity about ~10<sup>4</sup> higher than gas-filled detectors
- Helps in miniaturizing radiation monitoring instruments
- Ge(Li) used for gamma ray spectroscopy



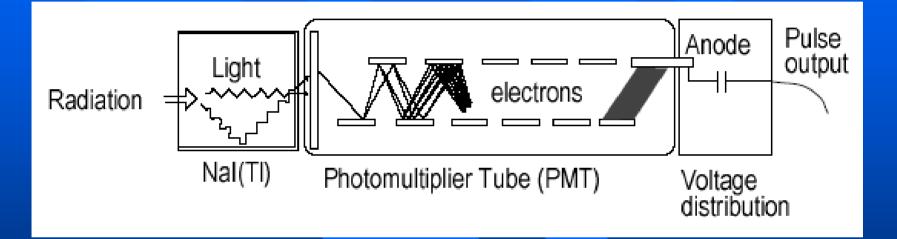


#### Scintillation Detectors

- Based on scintillation (light emission)
   Phosphor materials
  - chemical substance that exhibits fluorescence when excited by radiation (UV, X-ray, electron)
  - amount of visible light is proportional to the amount of excitation energy.
  - Includes NaI(TI), CsI(TI)

 NaI(Tl) coupled to a photomultiplier tube (PMT) often used for survey meters

#### Scintillation Detectors



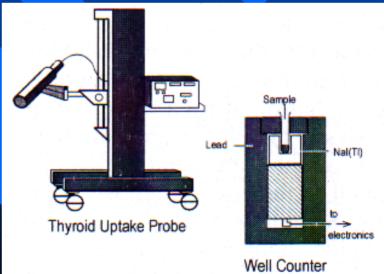
Solid crystal inside the probe
Electron excitation within the solid
Light emission and multiplication
Creates a current

# Scintillation Detectors

**Operation of NaI Detector** 

- Beta radiations cannot be detected.
- Efficiency is high (20%) for gamma contamination
- Need to use in low dose rate (<500 cpm).</p>
- Will readily see Bremsstrahlung radiation from P-32

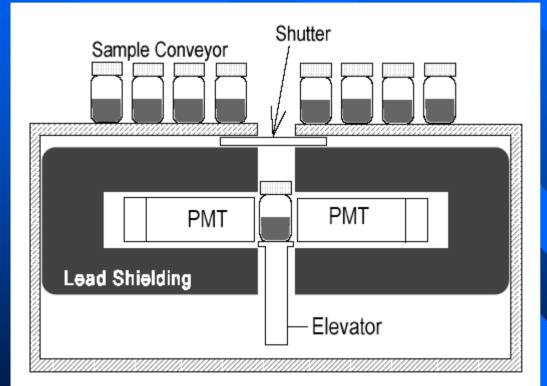




### Liquid Scintillation Detector

- Detects low energy beta particles (H-3, C-14)
- Use scintillation "cocktail" (organic solvent, fluorescent solute, dissolution agents) mixed with radioactive sample
- "Cocktail" absorbs radionuclide energy and emits light
- Light is collected by PMT
- All isotopes can be detected with high efficiencies (minimum 60%)

# Liquid Scintillation Detector



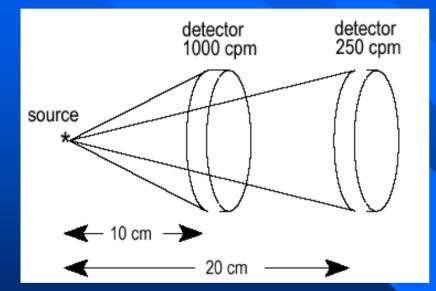


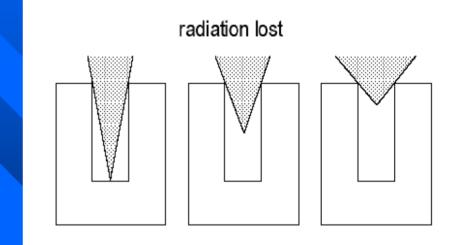
#### Automatic LS sample counter

### Detected vs Emitted Radiation

- Factors affecting measurement :
  - Counting Geometry
     » 1/r<sup>2</sup> law
  - Detector efficiency
     » Physical characteristics of detector
  - Deadtime
    - » Electronic limitation
  - Background radiation
     Natural bkg interfere with measurement
  - Counting statistics
    - » Radioactive decay is a random process

### **Detected vs** Emitted Radiation





1/r<sup>2</sup> law : Doubling
the distance
decreases radiation
detected by 4

Radiation losses vs source position in a well counter

#### Personal Dose Monitors

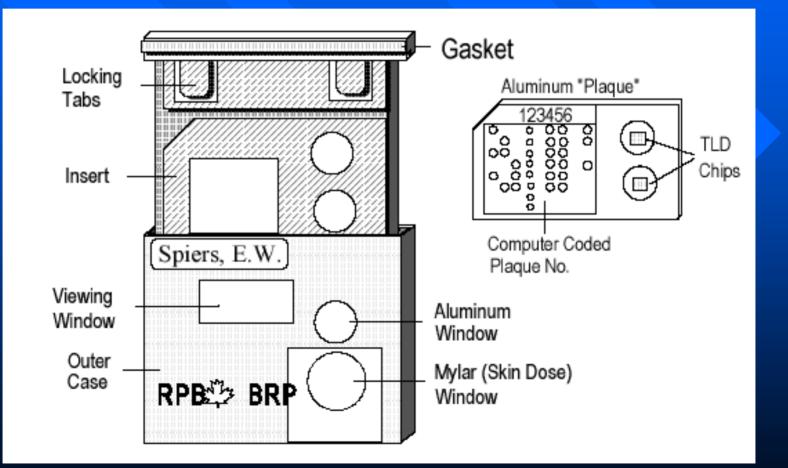
- Measures radiation doses received by individuals working with radiation
- Verify effectiveness of radiation control practices in the workplace
- Provide information in case of accidental exposure
- Most widely used are
  - TLD badges
  - Film badges
  - Direct reading monitors (EPD)

#### TLD chips

- Radiation => electrons are excited and trapped to certain energy levels in the material
- Light is released by heating
- Light is proportional to the amount of radiation absorbed by the TLD

#### TLD Badges

- Set of TLD chips in plastic holders with filters
- 2 radiation windows
  - » Thin Mylar window (7 mg/cm<sup>2</sup>) => skin dose
  - » Aluminum window (2 mm) => total body dose
- Send every 3 months to National Dosimetry Service for analysis and report

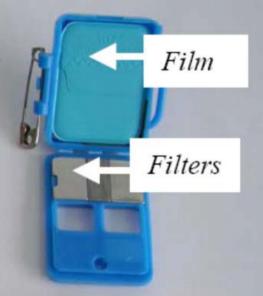


- Always wear your own all times while working
- Wear TLD's on chest area, UNDER protective clothing
- Wear extremity ring on most used hand, chip facing source
- Store away from radioactive sources
- DO NOT tamper with TLD chip or remove from holder
- Avoid subjecting to sunlight, high temperatures or wetness



# Film Badges

Alternative to TLD badge systems Photographic film covered by a set of filters Filters allow an assessment of the type of radiation involved Badges are returned to the monitoring service for development of the films Reading of the radiation dose from the optical density (opacity) of the film



#### **Radiation Detection**

#### Exposure

measure of: DOSE RATE instrument: dose rate meter

measures in dose per unit time (e.g. Sv/hr, rem /hr) Contamination Monitoring measure of: ACTIVITY instrument: contamination meter measures in counts per unit time (e.g.CPS, CPM)

#### Direct

monitoring of surface instrument: portable meter detects fixed & loose

#### Indirect

monitoring of sample (wipe) instrument: Liq. Scintillation or Gamma Counter detects loose contamination

#### Portable Instruments

At MUHC, portable survey instruments are used primarily for the detection of CONTAMINATION - the internal hazard.

Instruments can be used to detect the presence of RADIATION - the external hazard, but cannot be used to accurately quantify it

# Approximate detection efficiencies for some common radionuclides and detectors.

Radionuclide	LSC <sup>1</sup>	Pancake GM <sup>2</sup>	NaI(Tl) Meter <sup>3</sup>	NaI(Tl) Well <sup>4</sup>
H-3	20%	na <sup>5</sup>	na <sup>5</sup>	na <sup>5</sup>
C-14, S-35, P-33	50%	10%	na <sup>5</sup>	na <sup>5</sup>
Cr-51, Co-57, Tc-99m, I-125	30%	1%	50%	50%
P-32	100%	50%	na <sup>5</sup>	na <sup>5</sup>

LSC<sup>1</sup> : Liquid Scintillation Counter

PancakeGM<sup>2</sup> : Hand-held survey meter with pancake GM detector

NaI(Tl)<sup>3</sup> : Hand-held survey meter with well-type NaI(Tl) crystal

NaI(Tl)<sup>4</sup> : Multichannel analyzer with well-type NaI(Tl) crystal

na<sup>5</sup>: not applicable for this group of radionuclides

http://www.stanford.edu/dept/EHS/prod/researchlab/radlaser/manual/part3/surveys.htm

# MODULE 5

Transport and handling of Radioactive Packages

#### Transport of Dangerous Goods

Transport Dangerous Goods (TDG) Directorate



-focal point for the national program to promote public safety during TDG

-major source of regulatory development, information and guidance on dangerous goods transport for the public, industry and government employees.

#### Transport of Dangerous Goods

No person shall handle, offer for transport or transport DG unless trained or work under supervision of a trained person

Person is trained if employer is satisfied the employee has received adequate instructions for the handling of DG

Employer must ensure that trained persons have certificate of training (valid for 3 years)

Employee required to produce the TDG certificate if requested by the an inspector

### Classification of Dangerous Goods

\* \* \*

#### Class 1: Explosives



#### Class 3: Flammable liquids

### Classification of Dangerous Goods

Class 4: Flammable Solids



0

Class 5: Oxidizing Substances and Organic Peroxides

Class 6: Toxic and infectious substances







### Classification of Dangerous Goods

Class 7: Radioactive Materials







#### Class 8: Corrosives

8

Class 9: Miscellaneous Products, Substances or Organisms



### Class 7: Radioactive Materials

Material emitting hazardous radiation (radioactive particles)

-uranium hexafluoride -Thorium -Iridium -Tritium ...



### Class 7: Radioactive Materials

Radioactive packages may be shipped as:

- Excepted packages
- Industrial packages (for LSA material)
- Type A packages
- Type B (U) packages
- Type B (M) packages
- Type C packages

#### Marks and Labels

When handling or receiving radioactive materials the following proper shipping names and UN numbers may be observed (partial list only):

UN 2910 Radioactive material, excepted package, instruments

UN 2910 Radioactive material, excepted package, articles

UN 2910 Radioactive material, excepted package, limited quantity of material

UN 2910 Radioactive material, excepted package, empty packaging

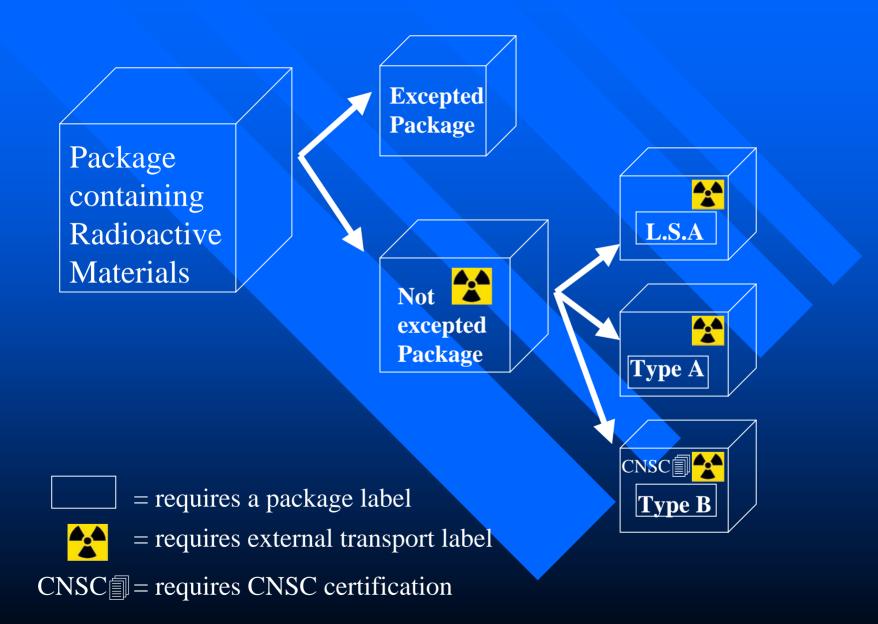
UN 2912 Radioactive material, low specific activity (LSA)

UN 2915 Radioactive material, Type A package, non-special form, non fissile or fissile excepted Class 7: Radioactive Materials - EXCEPTED PACKAGES

- Contains very low activities, very low hazard (ex. Calibration source sent to a high school)
- No special marking (could be a small cardboard box)
- The safety mark radioactive must be visible on opening the package
- All other packages must be categorized by radiation level and display the corresponding radiation warning labels

Class 7: Radioactive Materials TYPE A PACKAGES

- Contains medium activities of radioactive materials (ex. Radioisotope shipped to Hospital)
- Hazard in event of accident controlled by the limits and amount shipped
- Package may be metal, plywood, cardboard box or drum + foam inserts
- Package will have labels
- Transport vehicle requires placarding



### Marks and Labels

Radioactive material, type A package, UN 2915 Non special form, non-fissile or fissile exepted

**To**:



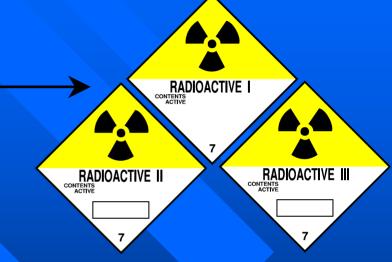
From : **ABC Chemical Co. MUHC** Buffalo, NY USA

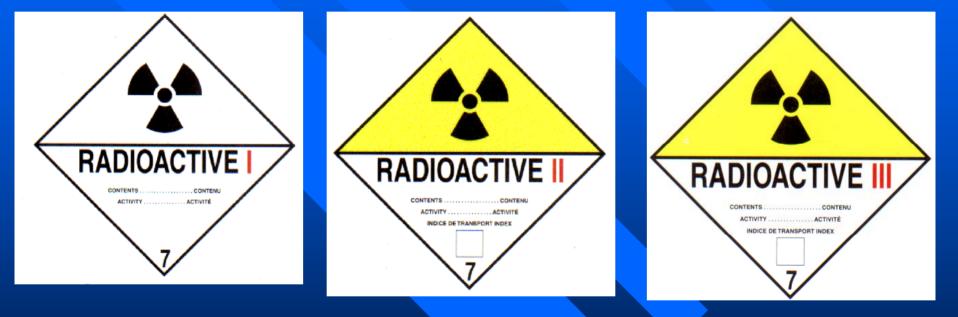
2450 Beach Road Montreal General Hospital. 1650 ave Cedar Montreal, PQ, Canada

### Vehicle Placarding

#### Transportation Placards And Labels







Category I - WHITE

- $\cdot$  < 5µSv/h at external surface
- Transport index < 1</li>

Category II - YELLOW

- Activity > excepted activity
   Activity > excepted activity
  - < 500 μSv/h at external surface
  - Transport index < 1</li>

Category III - YELLOW

- Activity > excepted activity
- $\cdot$  < 2000  $\mu$ Sv/h at external surface
- Transport index < 10</li>

# Category I No more 5 µSv/h (surface)



#### Category II

# - More then 5 $\mu\text{Sv}$ but no more than 0.5 mSv/h (surface)



Category III

 More than 0.5 mSv/h, but no more than 2 mSv/h (surface)



## Class 7 : Categories

# Category III (exclusive use) More than 2 mSv/h, but no more than 10 mSv/h



## Class 7 : Labels

Contents Activity (in Bq) Transport Index (T.I.) White I = no transport index <u>Yellow</u> II = TI between 0-1 Yellow III= TI between 1-10



T.I. = (dose rate in mSv/h) x 100 @ 1 meter

## **Receipt of Radioactive Material**

- Wear a Lab coat and disposable gloves
   Monitor radiation field
- Place in fume hood if volatile material
- Open outer package, check for possible damage
- Remove inner package, wipe test container
- Verify the radioisotope, activity ... with information on waybill
- Report any anomalies to the supervisor or RSO

See INFO-0426/rev1

Shipper AMERSHAM BIOSCIENCE 3350 N RIDGE AVE ARLINGTON HEIGHTS IL 60004			Air Waybill No. 4993 2713 2710 Page I of 1 Pages						
					Shippers Reference Number 959740				
687 PINE MOLECU	AVENUE LAR ONCO AL, PQ HE	WE: DLOC	ST 3Y GRC	UP					
Two completed and signed copies of this declaration must be handed to operator					WARNIN Failure to	comply in a	all respe	ects with the a	pplicable
TRANSPORT DETAIL			Airport of Departure		Dangerous Goods Regulations may be in breach of the applicable law, subject to legal penalties. This Declaration must not, in any circumstances, be completed and/or signed by a consolidator, a forward				
(delete non-applicable) PASSENGER AND	CARGO		СН	ICAGO	or an IAT	A cargo age	ont		
CARGO AIRCRAFT	AIRCRAFT ON	-		5	1	type: (delete 1	on-appli		-
Airport of Destination	n					NOASTIVE		RADIOACTIV	
NATURE AND QUAN	TITY OF DAI	NGER	OUS GO	ODS (see S	ab-section 8.1	of IATA Da	gerous C	Goods Regulation	ns)
Dangerou	s Goods Id	oods Identification			Ougntity and type of		Pocking	Authorizatio	
Proper Shipping Name	Class or Divison	UN ID I		Packing Group	Subsi- dairy Risk	Packing		Instructions	
RADIOACTIVE MATERIAL TYPE A PACKAGE	7	UNZ	1915			PHOSPHOR Liquid Organic Co		YELLOW T.I. 0.1	
								DIMENSIONS	
								26 x 17 x 12CM	
				Tax I Data Data Distance		ALL PACKED I 1 TYPE A PACI 519.468 M (14.040 T	KAGE VBq		
Additional Handling	Informatio	n IC	CAO / IA	<u> </u> .TA	J				
					R MEDICAL	TREATMEN	JT.		
		_				nergency sonusci		800-584	-9333
I hereby declare that the of consignment are fully and described above by prope me classified, packodyma are in all respects in her remport by air according	d accurately er shipping nan urked and labele proper conditio	ne and cd, and n for ple					Name/ JEFF M Place a CHIC/ Signa	GO AM	y EXP CLEF 1/25/2002

Shipper AMERSI 3350 N B ARLING	ł	Air Waybill No. 4993 2713 2710 Page 1 of 1 Pages Shippers Reference Number 959740					
687 PINE MOLECU MONTRE CANADA	VICTORIA H AVENUE N ILAR ONCOL IAL, PQ HIA	VEST .OGY GRO 1A1		WARNIN	0		
Two completed and sign handed to operator TRANSPORT DETAI This shipment is within th limitations prescribed for: (delete non-septicable)	Airp Depa	Airport of Departure CHICAGO		G comply in all res s Goods Regulat law, subject to I in must not, in an and/or signed b A cargo agent	ions may be in egal penalties iv circumstan	, breach of th This ces, be	
PASSENGER AND CARGO AIRCRAFT	CARGO AIRCRAFT ONLY		10.100		ype: (delete non-2p)	blicable )	
Airport of Destination					NOACTIVE	RADIOACT	
		ierous go	IODS (see Su	b-section 8.1	of IATA Dangerous	Goods Regulati	ions)
Dangerous Goods Identification					Quantity and type of	Packing	
Proper Shipping Name	Class 07 Divison	UN or ID No.	Pocking Group	Subsi- dairy Risk	Packing	Instructions	

	Class 07	lentification UN or	Pocking	Subsi-	Quentity and type of Packing	Packing Instructions	Authorizado
Proper Shipping Name	Divison	ID No.	Group	dairy Risk			
RADIOACTIVE MATERIAL TYPE A PACKAGE	7	UNI2915			PHOSPHORUS-32 Liquid Organic Compound	YELLOWI T.I. <b>O.I</b>	
						DIMENSIONS	
						26 x 17 x 12CM	
					ALL PACKED IN 1 TYPE A PACKAGE 519.468 MBq (14.040 mCi)		
Additional Handling							
				MEDICAL	TREATMENT		
					marganey contact Tel. No	800-584	9333
I hereby declare that the consignment are fully an described above by prop- are classified, packed,ma are in all respects in the transport by air according international and Nation	d accurately or shipping no arked and labo proper conditions to the applications	me and led, and on for ible			JEFF A Place CHICA	AGO AM	CLEF 1/25/2002

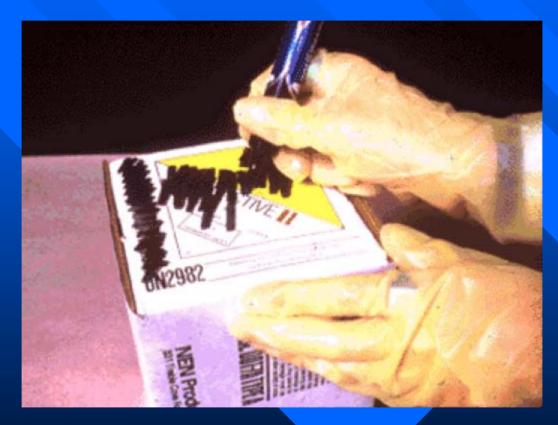
	SHIPPER'S DECLAR	ATION FOR DANGEROUS G	OODS (Provide at lea	ast two copies to the airlin	e.)				
[	Shipper SerkinElmer Life )549 Albany Stree Boston, MA 02118		Air Waybill No. 81468640 374 Page 1 of 1 Pages Shipper's Reference Number (optional) WARNING						
	Consignee HDSP ROYAL VICT UNIV CLINIC 687 AVE PINS D MONTREAL PQ CANADA H3A Two.completed and signed cop	181							
Ļ	be handed to the operator		Failure to comply in al	Failure to comply in all respects with the applicable					
	TRANSPORT DETAILS This shipment is within the limitations prescribed for: (delate non-applicable) PASSENGER AND CARGO AIRCRAFT ONLY	Airport of Departure BOSTON	Dangerous Goods Regulations may be in breach of the applicable law, subject to legal penalties. This Declaration must not, in any circumstances, be completed and/or signed by a consolidator, a forwarder or an IATA cargo agent.						
	Airport of Destination:	MONTREAL	Shipment type: (delete non-applicabl						
T	Radioactive mate /8-35,Liquid/Salt	rial,Type A pack ,1 Type A packag	age,7,UN2915// e x0.310762GBq//I-W	/ hite//#05102789	3.2				
					i				
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	Additional Handling Informatio	n							
	Entries comply w research having	dith IATA/ICAO direct applicati	This Shipment is on to HUMAN medical 24 hr. Emergency Contect Tel. No	welfare.	medical				
F	accurately described a	above by the proper marked and labelled/p	onsignment are fully and shipping name and are lacarded, and are in all according to applicable	Place and Dale	NS EXP REP TON, Ma APR-2003				

Air Waybill No. 8146864Ø 374 Page 1 of 1 Pages Shipper's Reference Number
WARNING
Failure to comply in all respects with the applicable Dangerous Goods Regulations may be in breach of the applicable law, subject to legal penalties. This Declaration must not, in any circumstances, be completed and/or signed by a consolidator, a forwarder or an IATA cargo agent.
Shipment type: (delete non-applicable)
NON-RADIDAQTIVE RADIOACTIVE

and all other required information.

Radioactive material, Type A package, 7, UN2915// N/-S-35,Liquid/Salt,1 Type A package x0.3107626Bg//I-White//#051027893 2 Additional Handling Information Entries comply with IATA/ICAO ---- This Shipment is to be used in medical research having direct application to HUMAN medical welfare. (703)527-38A7 24 hr. Emergency Contact Tel. No. \_\_\_\_ Name/Title of Signatory hereby declare that the contents of this consignment are fully and EWAN STEPHENS EXP REP accurately described above by the proper shipping name and are BOSTON, Ma Place and Date classified, packaged, marked and labelled/placarded, and are in all 02-APR-2003 respects in proper condition for transport according to applicable Signature international and national governmental regulations. (see warning above)

## **Receipt of Radioactive Material**



Remove or deface all radiation warning symbols prior discarding the box in the normal waste stream

#### **Certificate of Training Transportation of Dangerous Goods**

Name of Employer:

McGill University Health Center (MUHC) 1650 avenue Cedar, Montreal, PQ, Canada

Name of Employee:

This certificate certifies that the employee named above has completed the training for the *Safe Handling and Transportation Practices, and the Characteristics of Dangerous Goods* (class 7, radioactive materials), in accordance with the requirements of the Transportation of Dangerous Goods Act and Regulations.

Certificate expires on:

Employee's signature: Supervisor's signature: RSO's signature:

## MODULE 6

## Working in Laboratories

#### CLASSIFICATION OF LABORATORY

Exemption Quantity (EQ)

The quantity, in becquerel (Bq), of a radionuclide below which no licence is required

Isotope	EQ (MBq)
H-3	1000
C-14	100
P-32	0.01
P-33	1
S-35	100
Cr-51	1

RADIOISOTOPE	EXEMPTION QUANTITIY (EQ)				
	MBq	μCi			
H-3	1000	27000			
P-32	0.01	0.27			
P-33	1.0	27			
S-35	100	2700			
I-125	1.0	27			
I-131	0.01	0.27			
C-14	100	2700			
Cr-51	1	27			

CLASSIFICATION OF LABORATORY Annual Limit of Intake (ALI)

The activity, in becquerel (Bq), of a radionuclide that will deliver an effective dose of 20 mSv after the radionuclide is taken into the body

Basic:5XALIIntermediate:5-50XALIHigh:50-500XALI

> 10,000 EQ: Requires written approval from CNSC

RADIOISOTOPE	BASIC	LEVEL	INTERMEDIATE LEVEL			
	MBq	mCi	MBq	mCi		
H-3	5000	135	50000	1350		
P-32	40	1.08	400	10.8		
P-33	400	10.8	4000	108.1		
S-35	130	3.5	1300	35.1		
I-125	5	0.135	50	1.35		
I-131	5	0.135	50	1.35		
C-14	170	4.59	1700	45.9		
Cr-51	2650	71.6	26500	716		

#### CLASSIFICATION OF RADIONUCLIDES

Class A: Na-22, Co-60, Zn-65, ... Class B: Fe-59, Rb-86, ... Class C: H-3, C-14, P-32, P-33, S-35, Ca-45, I-125, ...

#### Contamination Criteria

**Decommissioning** Criteria

Class A : 3 Bq/cm<sup>2</sup> Class B : 30 Bq/cm<sup>2</sup> Class C : 300 Bq/cm<sup>2</sup> Class A : 0.3 Bq/cm<sup>2</sup> Class B : 3 Bq/cm<sup>2</sup> Class C : 30 Bq/cm<sup>2</sup>

## **Basic** Level

Do not eat, drink, store food or smoke In case of spill, follow spill procedure Clearly identify work surfaces for handling nuclear substances Check all packages containing nuclear substances for damage upon receipt Store nuclear substances in a locked room or enclosure when not in use Monitor laboratory for removable contamination weekly and keep records

#### Radioisotope Lab Laboratoire de radioisotopes

#### Room: AA-123

Basic Élémentaire

**RESPONSIBLE / RESPONSABLE John Smith Local: AA-123 Tel: 12345** 

-0-

#### En cas d'urgence / In case of emergency:

Contactez la radioprotection via (36111 Locating) Contact Radiation Protection via (36111 Locating)

### Intermediate Level

#### Basic level Procedures plus :

Wear appropriate dosimeters at all times » Extremity dosimeter required to handle more than 50 MBq of P-32, Sr-90, Y-90, Sm-153, Re-186

After working with NS, monitor work area for contamination

#### RAYONNEMENT-DANGER-RADIATION

Room: AA-123

Radioisotope Lab Laboratoire de radioisotopes Intermédiaire

RESPONSIBLE / RESPONSABLE John Smith Local: AA-123 Tel: 12345

#### En cas d'urgence / In case of emergency:

Contactez la radioprotection via (36111 Locating) Contact Radiation Protection via (36111 Locating)

## High Level

#### Intermediate level Procedures plus :

Restrict access to authorized personnel
Work in fumehood when required by RSO
Wash hands regularly and monitor them for contamination

#### RAYONNEMENT-DANGER-RADIATION

Room: AA-123

Radioisotope Lab Laboratoire de radioisotopes HIGH LEVEL Niveau Supérieur

**RESPONSIBLE / RESPONSABLE** John Smith Local: AA-123 Tel: 12345

#### En cas d'urgence / In case of emergency:

Contactez la radioprotection via (36111 Locating) Contact Radiation Protection via (36111 Locating)

### LABELING AND SIGNS

Use the words

RAYONNEMENT DANGER RADIATION



If quantity is > 100 EQ or If effective dose rate > 20 µSv/h

## LABELING AND SIGNS

#### Frivolous Posting of signs

- No person shall post a radiation sign (trefoil) at a place where radiation, nuclear substances or prescribed equipment is NOT present
- For pencil, rulers, calculators, pipettes, etc ... use any signs BUT the trefoil



Happy face will do fine !



#### Personal Protective Equipment Required for using Radioisotopes



Full Length Dresses or Pants ONLY

Eye Protection.

Latex Gloves

Lab Coat

Complete Coverage Feet and Legs

- No Shorts

- No Half Shorts
- No Open-Toed Shoes
- No Mini-Skirts

PPE is worn on the body Primary purpose is to provide a barrier to radioactive materials or radiation

## Personal Dosimeters

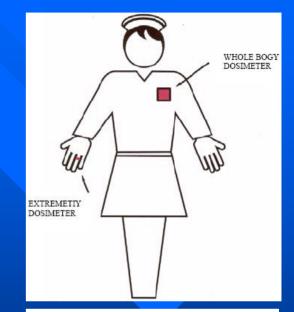
- Do not expose TLD to high temperature, water, sunlight
- Clip firmly between waist and neck
   Extremity TLDs (rings) facing the source
   If you loose your TLD, you should stop working until you receive a replacement
- Store in holder or rack when not in use
- Store in low radiation area away from light and heat

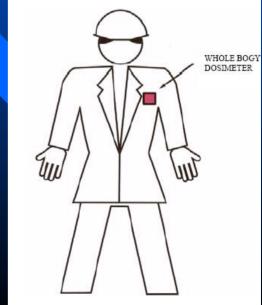


**Ring TLD** 



**Body TLD** 





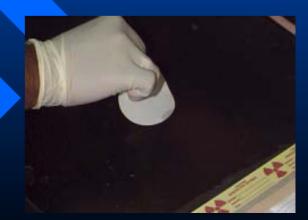
## Measuring Surface Contamination

#### Two Methods:

 The Direct Method Uses a pancake type probe, or thin walled Geiger tube to measure both fixed and loose contamination



 The Indirect Method Uses swipes to test for loose contamination



## The Direct Method

1.Select a slow response time and measure the background well away from the surface to be monitored

- 2. Select a fast response time, and pass the detector slowly over the surface at a distance of about 1 cm
- 3. Hold the probe over the detected area long enough to obtain a steady reading
- 4. Subtract the background rate from this reading and compare this result to the count rate corresponding to 1 CCL



## Pro's and Con's

#### Disadvantages:

- It is impossible to make a direct measurement of surface contamination when a high gamma field is present
- will not distinguish between loose and fixed contamination

#### Advantages:

- Measures the radiation level from both fixed and removable contamination

## The Indirect method

- 1. Select the surface to be swiped
- 2. Identify the swipe paper to be used
- 3. Swipe an area of about 100 cm<sup>2</sup> using light pressure
- 4. Put each swipe in separate envelope to prevent cross contamination
- 5. Measure the background count
- 6. Measure the activity of the swipe
- 7. The difference between the two readings is due to radioactivity picked up by the swipe



## Pro's and Con's

#### Disadvantages:

- Only 10% of the contamination will be picked up
- If the surface is wet the swipe needs to dry first
- Heavily contaminated swipes can lead to dead time losses in the detector

#### Advantages:

- Swipes measure loose contamination only
- They can be taken to a low background location to be measured

## RAD TAPE \* RADIOACTIVE \* RADIOACTIVE

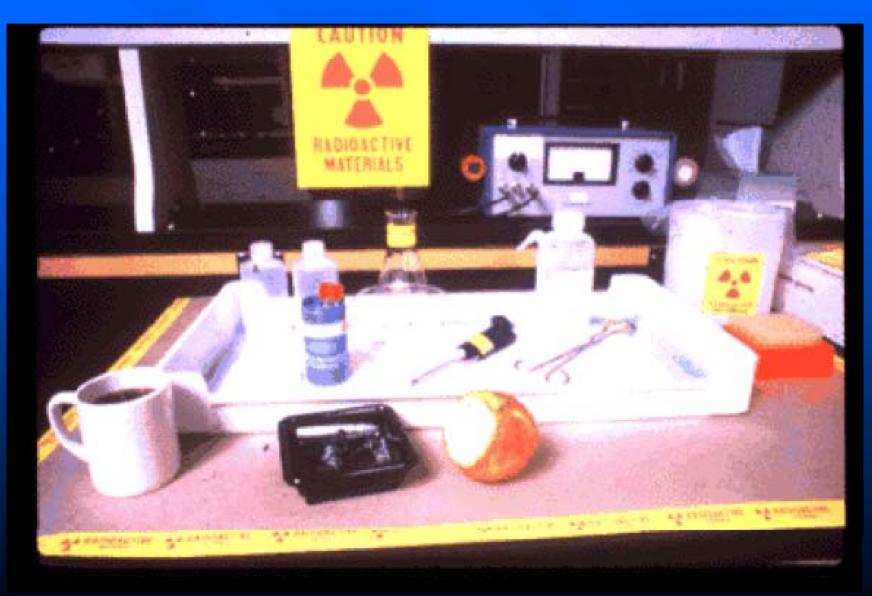


Work area delineated with rad tape

Use shield if available (ALARA)

Monitor for contamination when finished

## Find the Errors



## Find the Errors



### What to do to protect yourself....

#### Basic Radiation Safety Principles for External Radiation



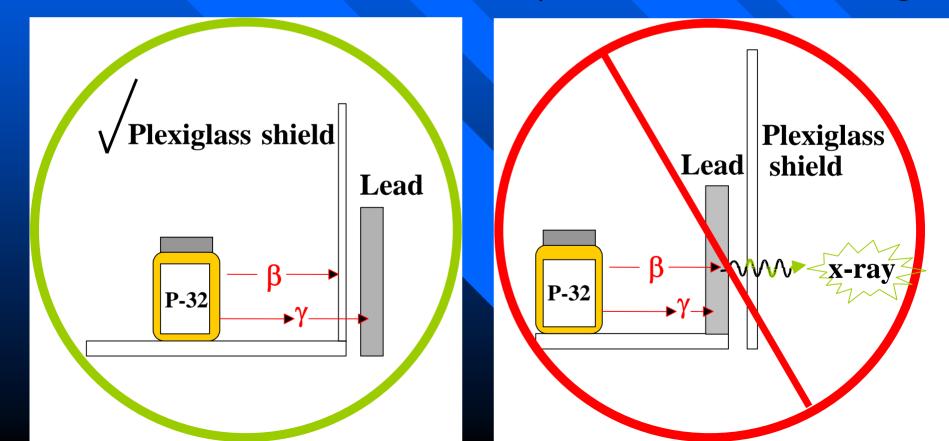


Work behind plexiglass shield when manipulation beta emitters

## Shielding Beta Radiation

Low density materials (ie. plexiglass) must be used to shield beta radiation because...

Interaction with dense materials produces Bremsstrahlung



Ordering radioactive material

- Only permit holders or authorized personnel can order radioisotopes
- Register new vials on the McGill Radioisotope Tracking System (RTS)

Inventory and disposal

 2 methods
 RTS (entered daily)
 Paper inventory (daily) before filing to RTS

Acquisition, control and storage

 Only permit holders or authorized personnel can order radioactive materials

 must be entered and tracked on the McGill Radioisotope Tracking System (RTS)

 must be tracked after each use on RTS system (new vial record)

 Byproducts created during handling which requires storage must be tracked also and labeled (user name, radionuclide, activity, date)



- Waste containers are also tracked by the RTS system
  - 1 isotope per container
  - Separate containers for liquid, solid, LSV

### Alternative Tracking (Forms)

- Running Log
- By-product form
- waste inventory form
- Forms
  - Must include the ID numbers assigned by RTS
  - Must be posted on the storage unit
  - Information must be entered on the RTS system every week

### **Radioisotope Running Log Form**

Radioisotope & Product:	Date of Measured Activity:	Volume:	
Activity:	Vial ID:	Date of Reception:	

Date	User	Activity Removed (uCi)	Volume Removed (ul)	Activity Remaining (uCi)	By Product ID	By Product RTS code	Liquid Waste (uCi)	Solid Waste (uCi)	Date of Disposal

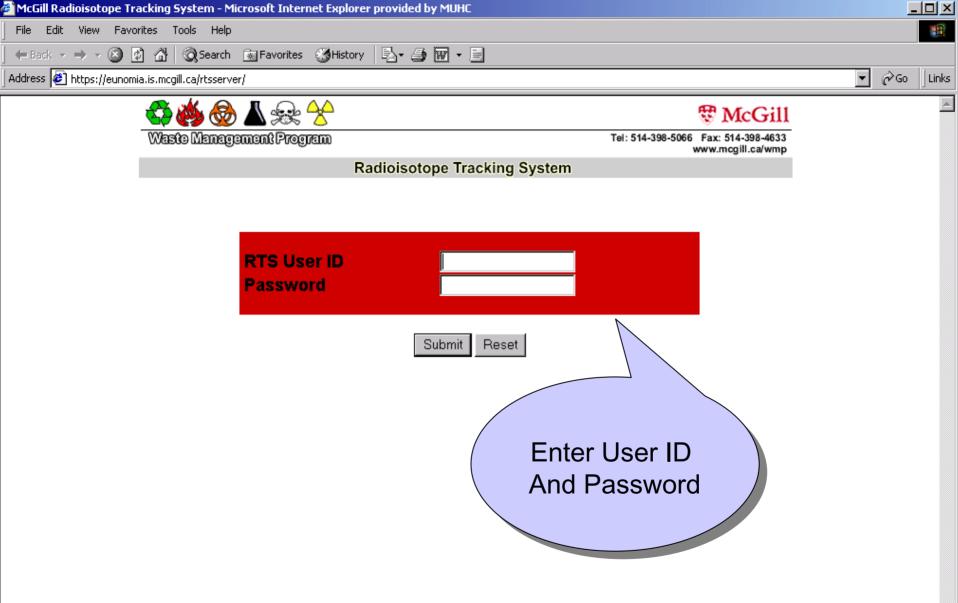
Disposed by: \_\_\_\_\_

Date of disposal of vial: \_\_\_\_\_ Activity disposed: \_\_\_\_\_

# McGill Radioisotope Tracking System (RTS)

Web application to manage the use of radioisotope and the disposal of waste.
Assigns a unique ID to vials
Tracks the usage while calculating decay
Enables the disposal through the proper waste containers

View the RTS web course online at : http://www.mcgill.ca/eso/training/presentations/#1





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🐯 McGill

Tel: 514-398-5066 Fax: 514-398-4633 www.mcgill.ca/wmp

**Radioisotope Tracking System** 

Licence Holder	Licence Holder Inventory				
	Radioisotope Vials	In Progress	Waste Containers		
Licence #: 5-0038-03 ROYAL VICTORIA HOSPITAL MS MAUREEN MCQUEEN RADIATION PROTECTION 842-1231 EXT 36133					





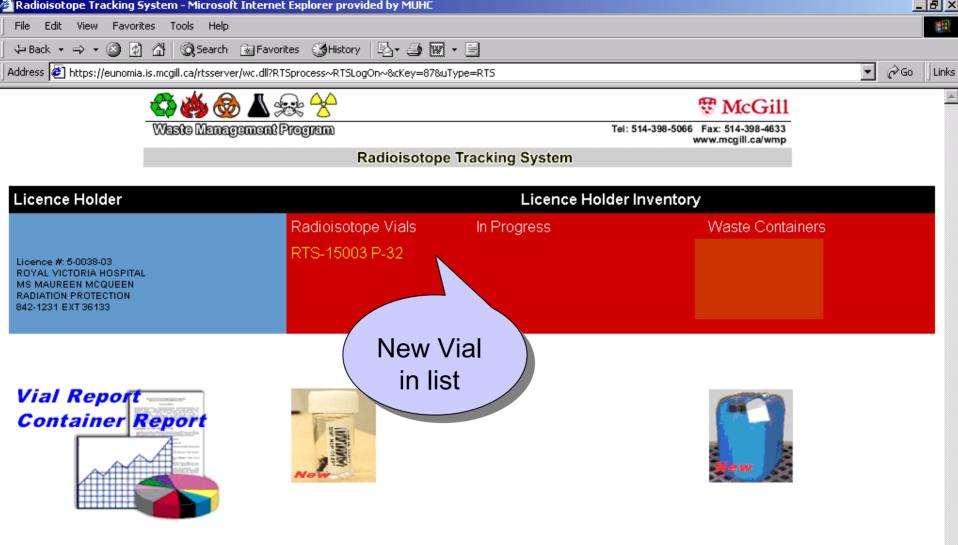




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842-1231 EXT 36133	Supplier	muhc					
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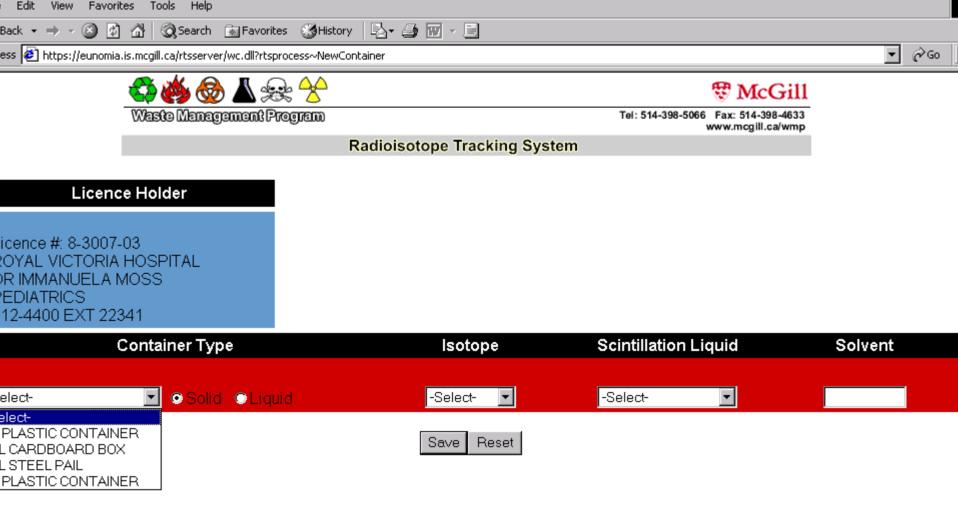


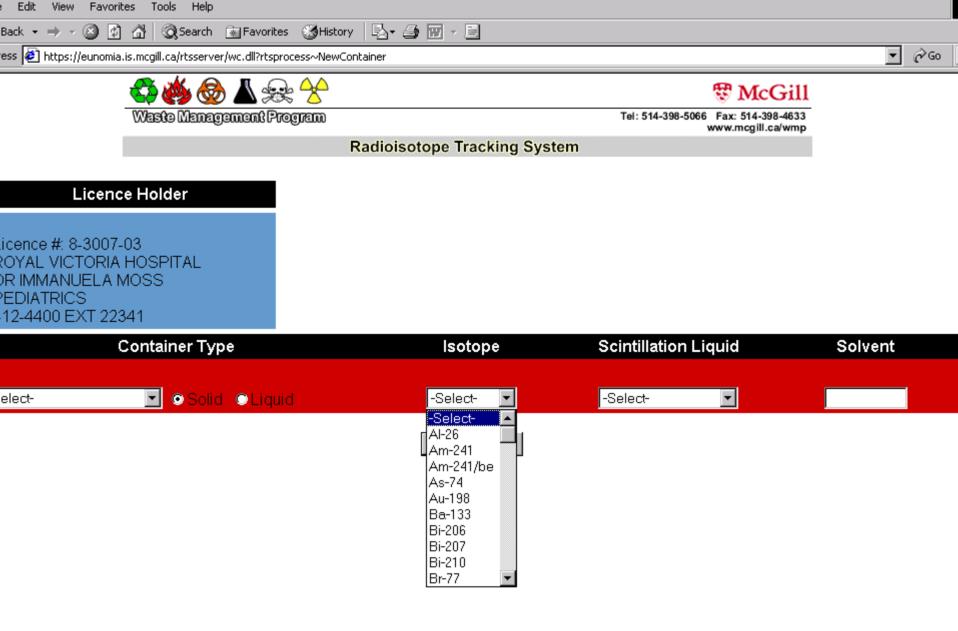


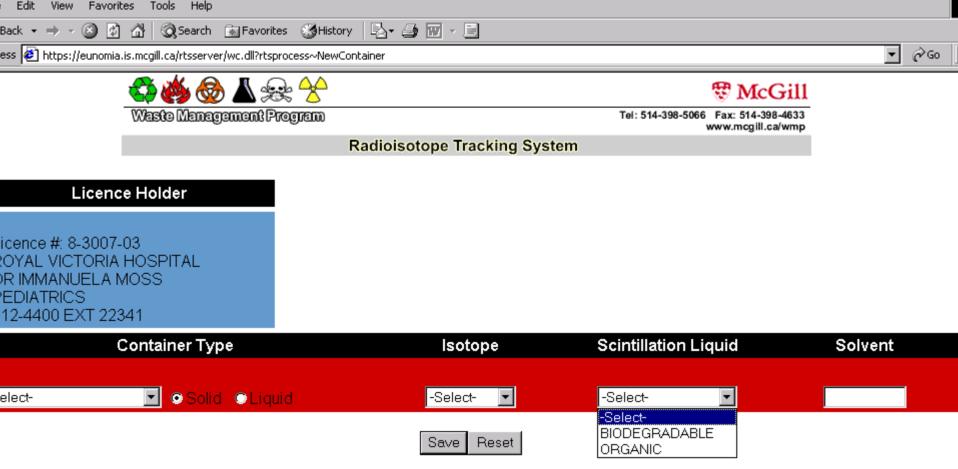








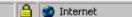




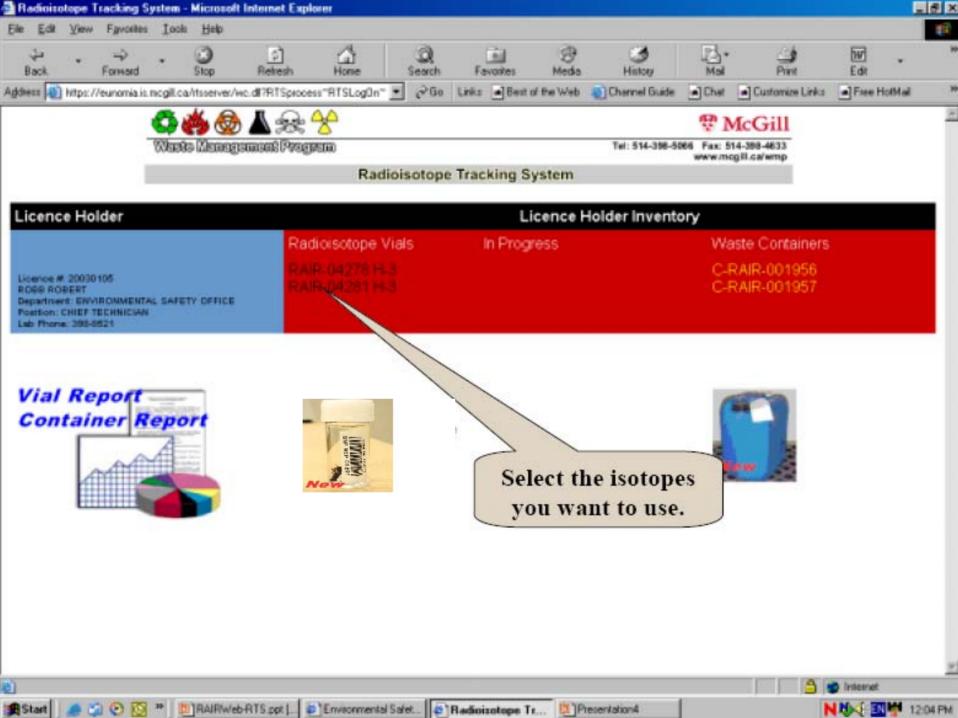
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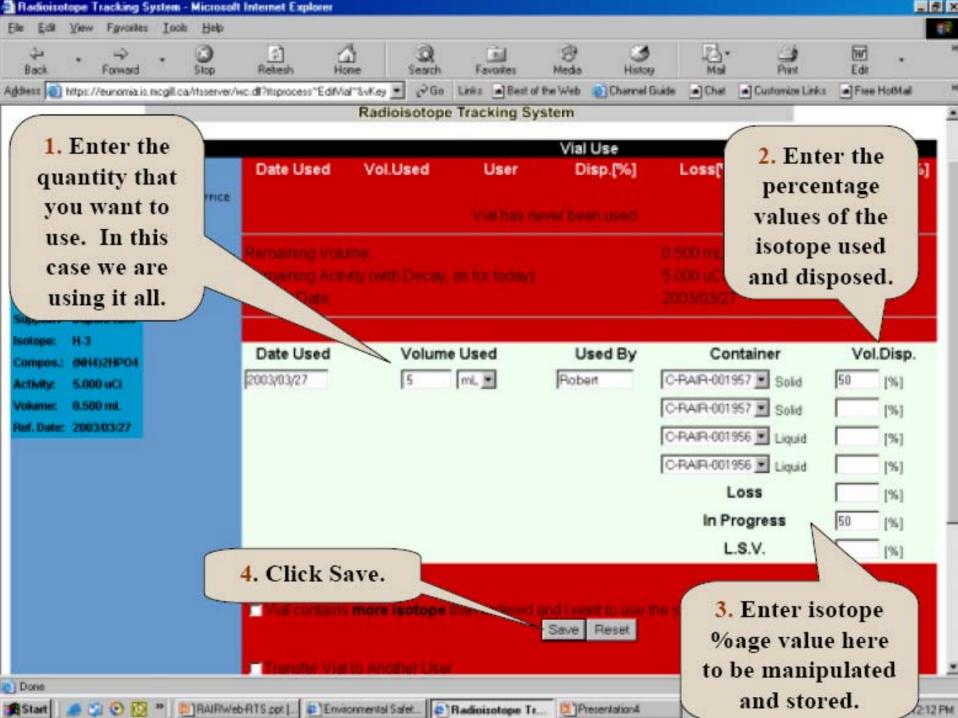


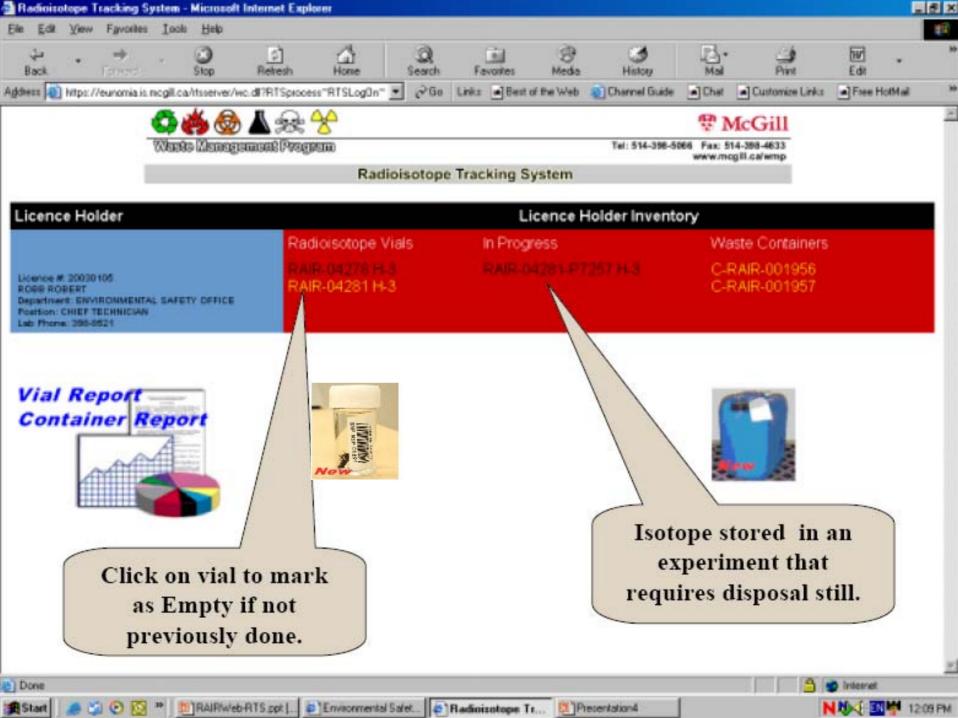
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RTS-00002	P-33	2002/03/15	2002/03/20	60.000	100.000	20.000	8.520 N	lo
		Date Use	Vol Used	User	Disposed[%]	Loss[%]	Progress[%]	
		2002/03/23	30.000 mL	CRIS	100.00	0.00	C	).00
		2002/03/25	10.000 mL	CRIS	100.00	0.00		0.00
Vial #	lsotope	Ref. Date	Rcvd. Date	Vol.Init[mL]	Activ.Init	Vol.Curr.[mL]	Act.Curr. Em [uCi]	pty
RTS-00017	Bi-210	2002/04/26	2002/04/26	45.000	200.000	45.000	-9.000 N	lo

TOTAL ACTIVITY IN = 400.000 uCi TOTAL VOLUME IN = 155.000 mL TOTAL ACTIVITY LEFT = 37.345 uCi TOTAL VOLUME LEFT = 115.000 mL

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DR IMMANUELA MOSS		2003/04/01	7.511	7.511	RTS-04003	A LAFERRIERE	
PEDIATRICS		2003/12/08	11.563	0.811	RTS-09184	A LAFERRIERE	
412-4400 EXT 22341		2004/05/07	13.462	2.138	RTS-12211	A LAFERRIERE	
Container No.C-RTS-001985 Isotope I-125 Cont. Type 4L PLASTIC CON Form Liquid Solvent Entered 2003/03/29 Last Used 2003/03/29	TAINER	Total Activity (without Activity to Date (with I	Decay) = 2.428	iner is Full to Obtain La	abel Information		



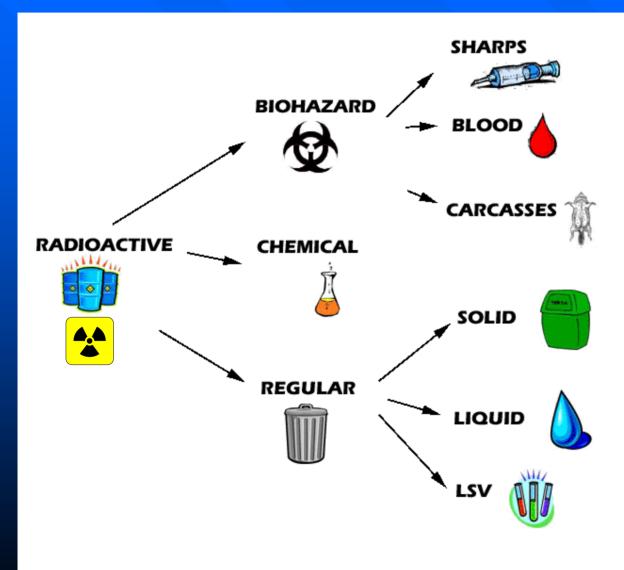
# Waste Management



## Radioactive Waste Disposal

- Segregate waste by form (solid, liquid) and isotope
- A running total amount of activity and isotope in waste must be kept on container - replace as required
- Label waste as well as record in RTS inventory

### Waste Disposal Guidelines







# Regular Radioactive Waste



4 L white plastic container 1 L clear plastic container

Both used for liquid or for solids (no mixing of solids & liquids).



Metal pail (20 litre) used for solid and for liquid scintillation vials (LSVs).

Cardboard box (20 litre) used for solid waste

# Mixed Biomedical & Radioactive Waste

Disposal of biomedical waste is governed by the Regulation Respecting Biomedical Waste (Québec), and encompasses the following categories:

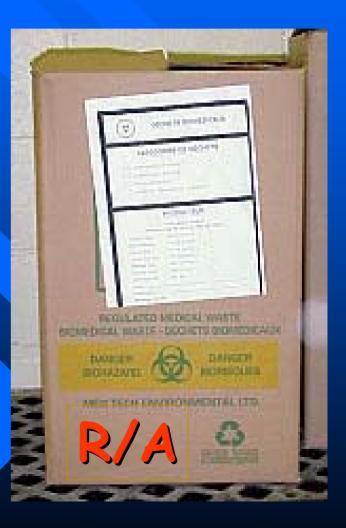
- human anatomical waste (body parts or organs),
- animal anatomical waste (carcasses, body parts, organs),
- non-anatomical waste, which includes:
- sharps which have contacted animal or human blood, biological fluids or tissues
- tissue or microbial cultures, and material contaminated by such cultures
- live vaccines
- containers or materials saturated with blood products.

# Mixed Biomedical & Radioactive Waste

For biohazard waste that are radioactive, <u>always use</u> <u>biohazard waste route</u> and write



in red on the box (as per McGill Waste Management instructions)



# LSV Waste Tag

- 1. Isotope / Isotopes
- 2. Activity
- 3. Date *measured*
- 4. # Permit
- 5. Initials





# Lead/Plastic Pigs

Lead pigs are recyclable. -Must be tested for contamination. Plastic pigs are not recyclable -may be presented as garbage -must be free of contamination and radioactive markings.



### Code brown: Internal spill or release of hazardous materials

### Call 55555

Department responsible: Occupational Health and Safety (OH&S)

### **Definition:**

Code brown is used to alert MUHC personnel of an internal hazardous spill of chemical and pharmaceutical products, radioisotopes and biohazardous materials, as well as gas leak

In the event of a chemical spill, the individual(s) who caused the spill or the first individual who noticed the spill or leak is responsible for prompt and proper communication with other occupants, his/her immediate supervisor, and locating.

# Spill Procedure

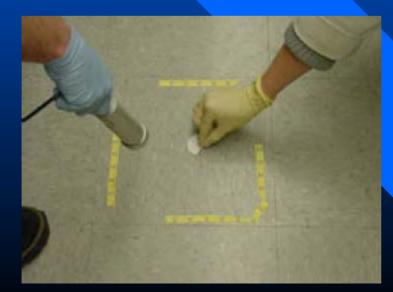
### MINOR SPILLS (< 100 EQ)

- Inform persons in immediate area, limit access
- Cover spill with absorbent material
- Clean up spill using absorbent paper and place in plastic bags
- Wipe test or survey for residual contamination as appropriate.
- Repeat until decontamination criteria is exceeded
- Check hands, clothing and shoes for contamination
- Package and label clean up materials for disposal.
- Record spill details and contamination monitoring (wipe tests) results.
- Adjust inventory and waste records as appropriate



### Assess Contamination (isotope, activity) Minor or major ?

### Clean up using absorbent pads



Survey, identify region

repeat cleaning + wipe tests as necessary

# Spill Procedure

### MAJOR SPILLS (> 100 EQ)

- Clear the area
- Leave fume hood running (if available)
- Close off area and post warning signs
- Notify RPS
- Decontaminate personnel by removing clothing and flushing skin with warm water and soap
- Follow procedure for minor spill
- Record names of person involved in spill
- Submit written report to RPS
- RPS will submit report to CNSC

# Appendix I

# Two Important Calculations

## CALCULATIONS

### TWO IMPORTANT CALCULATION:

1. Wipe test contamination level

2. Decay correction

## Wipe Test Contamination Level

$$CL(Bq/cm^{2}) = \frac{N(CPM) - Bkg(CPM)}{CE \times 60 \times A \times Weff}$$

- $CL = Contamination Level in Bq/cm^2$
- N = Total counts in CPM
- Bkg = Background counts in CPM
- CE = Detector efficiency (0-1)
- 60 = sec/min
- A = Area wiped (100 cm2)
- Weff = Wipe efficiency (0.1 wet; 0.01 dry)

## Wipe Test Contamination Level

### EXAMPLE :

- N = 500 CPMBkg = 300 CPM
- CE = 0.6
- A = 100 cm2Weff = 0.1

(Total counts)
(Bkg counts)
(Det efficiency)
(Area wiped)
(Wipe efficiency)

$$CL = \frac{500 - 300}{0.6 \times 60 \times 100 \times 0.1} = 0.55Bq / cm^2$$

## **Decay** Correction

$$A_f = \frac{A_0}{2^N}$$
;  $N = t / T_{1/2}$ 

 $\begin{array}{ll} A_0 &= Initial \ activity \\ A_f &= final \ activity \\ t &= time \ elapsed \\ T_{1/2} &= half-life \end{array}$ 

Note: 
$$A_f = A_0 \exp(-\ln(2^N))$$
  
=  $A_0 \exp(-N \times \ln(2))$   
=  $A_0 \exp(-0.693 \times t / T_{1/2})$   
 $A_f = A_0 \exp(-\lambda t)$  ( $\lambda = 0.693 / T_{1/2}$ )

## **Decay** Correction

### Example:

- 250 µCi of <sup>35</sup>S arrived on May 19, 2002
   100 µCi was removed and used the same day.
   The remaining amount was stored in a freezer for future use.
- On June 30, 2002 it is decided to repeat the experiment.
- **Q** :Does another order of <sup>35</sup>S have to be placed or is there enough remaining activity that the experiment may proceed?

## **Decay** Correction

Solution:

$$A_{f} = A_{o} \exp(-\lambda t)$$

 $\begin{array}{l} A_{\rm f} = {\rm activity \ at \ time\ "t"\ (?)} \\ A_{\rm 0} = {\rm activity\ at \ time\ zero\ (250 - 100 = 150\ \mu Ci\)} \\ t = {\rm elapsed\ time\ (42\ d\)} \\ \lambda = {\rm decay\ constant\ (0.693\ /\ 87.44\ d\ = 0.0079)} \end{array}$ 

 $A_{\rm f}$  = 150 exp( - 0.0079 x 42 ) = 107.52 µCi

\*\* SAVINGS \*\*

### Le centre universitaire de santé McGill (CUSM) McGill University Health Center (MUHC)

### Service de radioprotection Radiation Protection Service

Ceci certifie que : This is to certify that :

A réussi avec succès la formation du CUSM en radioprotection à l'intention des travailleurs du secteur des radioisotopes. Has successfully completed the MUHC radiation Safety Training for Radioisotope Workers

Date of training:

RSO's signature:

Service de la radioprotection Radiation Protection Service

## MUHC Radiation Protection Service

Christian Janicki RSO and manager x 43866
Malika Oussaid, RS Assistant x 36484

- Mohammed Lounis, RS Assistant x 44922

- 24h pager through Locating x 53333